

Iranian Neo-Tethyan magmas as a significant CO₂ source during the Middle Eocene Climate Optimum

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Highlights

- First measurements of volatile content in Eocene Neo-Tethyan lavas from Iran.
- 1.01×10^{19} g CO₂ (minimum estimate) were released from the studied magmas.
- The degassed CO₂ likely contributed to the Middle Eocene Climatic Optimum.

Abstract

CO₂ emissions from magmatic arcs can affect the atmosphere composition, thereby driving long-term global climate changes. Early Cenozoic climate trends are generally associated with changes in global silicate weathering related to Neo-Tethyan geodynamics, but the likely climatic effects of changes in degassing from Neo-Tethyan magmatic arcs have been poorly quantified. Here, we characterize the petrography and provide the first measures of the volatile content (CO₂, H₂O, F, Cl and S) of pre-eruptive melts based on glassy, bubble-bearing and reheated melt inclusions within plagioclase and clinopyroxene crystals in Early Cenozoic trachyandesites from the Alborz and Tabriz regions (Iran). CO₂ concentrations in these melt inclusions reach up to 6733 ppm, thus providing a minimum estimate of the total amount of CO₂ degassed from Iranian magmas during the middle-late Eocene of 1.01×10^{19} g CO₂, with a total C flux released of 0.306 (\pm 0.012) Mt C/yr, which is within the carbon imbalance predictions estimated based on other proxies. Our measures validate earlier hypotheses that magmatic CO₂ degassing from the targeted igneous provinces contributed to the Middle Eocene Climatic Optimum. Further measurements of the volatiles content of Neo-Tethyan magmas are thus critical to assess the drivers of Early Cenozoic climate trends and understand the global volatile cycling over geological timescales.

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1. Introduction

On timescales of millions to tens of millions of years, the geological carbon cycle controls how much carbon is stored into rocks and how much carbon is released into the ocean and atmosphere (Berner and Lasaga, 1989; Sundquist, 2013). Plate tectonics and magmatism are

42 thus linked to the evolution of climate and life, as volcanoes and magmatic provinces are
43 sources of carbon by releasing greenhouse gases [e.g., carbon dioxide (CO₂)], as well as H₂O,
44 sulfur, and halogens [e. g., F, Cl, Br] (Berner, 2003; Berner and Lasaga, 1989; Burton et al.,
45 2023; Cloetingh et al., 2023; Walker et al., 1981). CO₂ is one of the major components of
46 volcanic gases and evidence for higher carbon output was shown for arcs where carbonate
47 sediments subduct (Freundt et al., 2014) or where magma-carbonic rocks interactions occur
48 in the upper crust (Aiuppa et al., 2017). Understanding the long-term interactions between
49 climate and magmatism is also critical to locate and manage natural resources as well as
50 mitigate natural hazards, but the relationships between continental subduction/collision and
51 magmatic activity are, to date, particularly poorly constrained (Sternai, 2023).

52 During the Paleocene and early Eocene, approximately from 60 Ma to 50 Ma, a climate
53 warming trend is observed, followed by a 15 Myr period of cooling, between 50 and 35 Ma
54 (Beerling and Royer, 2011; Hansen et al., 2013; Zachos et al., 2001). The Eocene cooling trend
55 is classically ascribed to atmospheric CO₂ consumption by tectonic uplift, erosion and silicate
56 weathering that occurred along the Eurasian Margin in relation with the India-Eurasia collision
57 (Raymo and Ruddiman, 1992). Eocene cooling, however, is perturbed by a significant sharp
58 temperature increase anomaly at ~40.5 Ma lasting 500 kyr and referred to the Middle Eocene
59 Climatic Optimum (MECO) (Ma et al., 2024; Simon et al., 2008; Sluijs et al., 2013; van der
60 Boon et al., 2021). Although more than 15,000 km of magmatic arcs were active along the
61 convergent Neo-Tethyan plate margins throughout the Mesozoic and the Paleogene (Seton et
62 al., 2012; Sternai et al., 2019), the possibility that the considerable and variable amount of CO₂
63 sourced from these arcs into the ocean and atmosphere contributed to global climate was only
64 marginally explored thus far (Hoareau et al., 2015; van der Boon et al., 2021; Zhang et al.,
65 2023). Interestingly, magmatic flare-ups from the Neo-Tethyan arcs in Iran occurred between
66 ~50-35 Ma, peaking around 45-40 Ma (Alpaslan et al., 2004; Arslan and Aslan, 2006; Kazmin
67 et al., 1986; van der Boon et al., 2021; Verdel et al., 2011; Vincent et al., 2005), which overlaps
68 with the MECO. Previous authors suggested that intensified volcanic activity in Iran may have
69 contributed to the MECO (Bohaty et al., 2009; van der Boon et al., 2021). However, this
70 hypothesis remains largely untested by direct measurements, which are only possible through
71 melt inclusions (MI) to assess the pre-eruptive volatile content of melts (Esposito, 2021, and
72 references therein). Melt inclusions are melt droplets trapped in crystals during their growth
73 (Esposito, 2021; Metrich and Wallace, 2008, and references therein). The preserved pre-
74 eruptive volatile content of magmas can be used to assess the storage conditions of magmas
75 and estimate the CO₂ outgassing for magmatic systems where MI can be retrieved (Balcone-
76 Boissard et al., 2023; Esposito, 2021; Esposito et al., 2023, 2011; Hauri, 2002; Lowenstern,
77 1995; Metrich and Wallace, 2008; Putirka, 2008; Roedder, 1979; Rose-Koga et al., 2021;
78 Sobolev, 1996; Sobolev and Danyushevsky, 1994).

79 Here, we present the first measurements of CO₂ concentrations within MI in plagioclase
80 and clinopyroxene crystals from Eocene Iranian trachyandesites. These measures enable us to
81 constrain the pre-eruptive volatile content of Eocene Iranian lavas and provide the first volatile-
82 based estimate of the regional magmatic CO₂ emissions to evaluate the proposed Iranian
83 magmatic forcing on the MECO (De Leeuw, et al., 2007; Hilton, et al., 2002; Sano & Williams,
84 1996; Wallace, 2005).

85 2. Regional geology of Iran

86 Iran is located on the Arabia-Eurasia plate margin, an extensive region that is actively
87 deforming due to ongoing continental collision following the subduction of the Neo-Tethys
88 Ocean (Agard et al., 2005; Hatzfeld and Molnar, 2010; Vernant et al., 2004). The northward
89 subduction of oceanic lithosphere beneath Iran occurred at fairly constant rates, between 2 and
90 3 cm/yr, since ~56 Ma (McQuarrie et al., 2003). The timing of collision onset between Arabia
91 and Eurasia remains controversial; early works proposing from the Late Cretaceous to the
92 Miocene or even the uppermost Pliocene (Alavi, 1994; Berberian et al., 1982; Berberian, 1983;
93 Berberian and King, 1981; Falcon, 1974; Stöcklin and J., 1968), whereas more recent studies
94 seem to agree on a late Eocene-Oligocene collision initiation (Agard et al., 2011, 2005; Ballato
95 et al., 2011; Fakhari et al., 2008; Jolivet and Faccenna, 2000; Koshnaw et al., 2019; McQuarrie
96 and van Hinsbergen, 2013; Pirouz et al., 2017; Verdel et al., 2011; Vincent et al., 2005).
97 Regardless of its onset timing, the Arabia-Eurasia collision mainly led to the formation of the
98 Zagros, Caucasus, Alborz, Kopeh-Dag and Talesh mountains in Iran (Boutoux et al., 2021;
99 Hatzfeld and Molnar, 2010) (**Fig. 1**).

100 During the Paleogene, subduction of the Neo-Tethyan oceanic lithosphere beneath Iran
101 at rates of 2-3 cm/yr produced the ‘Andean-type’ Urumieh-Dokhtar arc in Central Iran,
102 extending over ~1500 km (Boutoux et al., 2021; Dewey et al., 1973; McQuarrie et al., 2003).
103 The Urumieh-Dokhtar belt is one of the three areas in Iran where Eocene calc-alkaline volcanic
104 rocks are preserved, along with the southern Alborz Mountains, and the Lut block of eastern
105 Iran, with an overall estimated volumes of magmatic material in the order of 10^6 km³ (Verdel
106 et al., 2011) (**Fig. 1**). The Alborz mountains extend from Talesh in the west, where they are
107 characterized by a back-arc system signature (Asiabanha and Foden, 2012; Golonka, 2004;
108 Vincent et al., 2005), to Kopet Dag in the east (van der Boon et al., 2017). Evidence of
109 submarine volcanism has been reported in the Alborz mountains where at least 3000- to 4000-
110 m-thick volcanic, volcanoclastic and sedimentary strata currently outcrop, as the Karaj
111 Formation where locally up to 8000 m of volcanoclastic sediments are present (Amidi et al.,
112 1984; Amidi and Michel, 1985; Ballato et al., 2013; Berberian and King, 1981; Förster et al.,
113 1972; Verdel et al., 2011). Previous works in this region also show mafic to intermediate
114 andesitic and dacitic lavas associated to a magmatic pulse in the Eocene (Amidi et al., 1984;
115 Amidi and Michel, 1985; Azizi and Jahangiri, 2008; Förster et al., 1972; Verdel et al., 2011),
116 preceded and followed by minor or no magmatism (Pearce et al., 1990). While previous studies
117 have estimated the pressure and temperature storage conditions of Oligocene magmas from
118 Urumieh-Dokhtar magmatic arc and Quaternary volcanic rocks (Babazadeh et al., 2021; Salehi
119 et al., 2020), the Eocene volcanics are less studied and their volatile content is currently
120 unconstrained. Our study focuses on Eocene volcanic rocks located along the northwestern
121 sector of the Iranian Plateau (north of the city of Tabriz) and the southern flank of the Alborz
122 mountains (Karaj Formation) (**Fig. 1**).

123 A widespread magmatic flare-up that delivered a large fraction of the total magmatic
124 production occurred throughout Iran in the Eocene, despite a relatively slow and steady rate of
125 subduction (Azizi and Moinevaziri, 2009; Ducea, 2001; Ducea and Barton, 2007; Verdel et al.,
126 2011). The cause of this flare-up has been long debated (Amidi et al., 1984; Amidi and Michel,
127 1985; Förster et al., 1972; Shafaii Moghadam et al., 2015). By combining new age constraints
128 and geochemical data from primitive mafic rocks emitted during and after the flare-up, Verdel

129 [et al. \(2011\)](#) proposed a model where Eocene extension and crustal thinning, accompanying
130 slab rollback generated decompression melting of a lithospheric mantle hydrated by slab-
131 derived fluids (see also [Shafaii Moghadam et al., 2015](#)). A similar dataset led [Rabiee et al.](#)
132 [\(2020\)](#) to propose that upwelling in the back-arc region generated asthenospheric melt, which
133 interacted with the metasomatized supra-subduction mantle wedge and/or the continental crust,
134 resulting in magma with heterogeneous characteristics.

135 [Van der Boon et al. \(2021\)](#) explored the link between the Iranian magmatic arc flare-up and
136 the MECO, and estimated the total amount of carbon released during the flare-up between
137 1052-12565 Pg C using the spatial extent of presently outcropping volcanic rocks as a base to
138 constrain the total volume of erupted volcanics. It should be noted, however, that this estimate
139 most likely represents minimum values because endorheic conditions persist in Central Iran,
140 where volcanic rocks are probably buried beneath a several-kilometer-thick sequence of late
141 Cenozoic marine and continental sediments. Nevertheless, this estimate matches the amount of
142 carbon released required to produce the MECO ([Henehan et al., 2020](#); [Sluijs et al., 2013](#); [van](#)
143 [der Boon et al., 2021](#); [van der Ploeg et al., 2018](#)). However, the CO₂ content of the magmas
144 associated with the Eocene magmatic flare-up in Iran was not directly quantified ([van der Boon](#)
145 [et al., 2021](#)). Hence, direct petrological measurements are required to verify the possible links
146 between the Eocene Iranian flare-up and the MECO.

147

148 **3. Material and methods**

149 The analyzed samples include three trachyandesitic samples from the NW sectors of the Iranian
150 Plateau, in the region of Tabriz and the southern Alborz Mountains, representing the back-arc
151 of the Eocene subduction system (**Fig. 1**; **Supplementary Fig. 1**; their coordinates are
152 provided in **Table 1**) ([Geological Survey and Mineral Exploration of Iran, 2006, 1995](#)).
153 Geological maps constrain the samples ages to the middle-late Eocene ([Geological Survey and](#)
154 [Mineral Exploration of Iran, 2006, 1995](#)). The samples have been cut for thin sections
155 (University of Milano-Bicocca and Petrolab, Sardegna), crushed for grains separation, and
156 grinded for powder (whole rock analyses). Minerals were separated using a Frantz magnetic
157 separator and handpicked under a stereomicroscope. The next steps of MI preparation as well
158 as Raman, SIMS, and microprobe analyses are presented below and in more details in the
159 **Supplementary Note 1**.

160 **3.1. Whole rocks**

161 Whole rock major-element compositions were analyzed using an inductively coupled plasma
162 optical emission spectrometer (ICP-OES) at Bureau Veritas Mineral Laboratories (Canada).
163 Oxides that were characterized are SiO₂, TiO₂, Al₂O₃, Fe₂O₃, MnO, CaO, Na₂O, K₂O and P₂O₅.
164 Internal standards from Bureau Veritas were used for calibration.

165 **3.2. Melt inclusions preparation**

166 MI bearing samples were prepared in the laboratory of the University of Milano-Bicocca, and
167 sample preparation details are reported in the **Supplementary Note 1**.

168 Crystals of plagioclase and clinopyroxene were mounted individually in Crystalbond 509 on a
169 heating plate. If the MI in the crystals were homogeneous, each crystal was polished again to
170 expose the targeted MI (**Supplementary Fig. 2**). If the MI were crystallized, heating
171 experiments were performed, and after the experiment, quenched MI were exposed to the
172 surface of the crystals by gently polishing them (**Supplementary Fig. 2-3**; see below). A
173 second extraction of the crystals from the Crystalbond was carried out to press the crystals with
174 the exposed MI in an indium mount for analysis of volatiles in the MI glasses to prevent epoxy
175 contamination driven by low-vacuum by Secondary Ion Mass Spectrometry (SIMS).

176 **3.3. Heating experiments**

177 Melt inclusions in the Iran trachyandesites (**Table 1**) required heating and quenching to obtain
178 a homogeneous glass, as MI are partially or totally crystallized in all plagioclase crystals and
179 some of the clinopyroxene crystals. Heating of the MI was necessary as analyzing crystallized
180 MI can generate misleading information about the original melt composition and volatile
181 content ([Esposito et al., 2023](#)). 33 heating experiments were conducted at University of Milano-
182 Bicocca (Italy) using a one-atmosphere Linkam TS1400XY heating stage ([Esposito et al.,](#)
183 [2012](#)). The heating stage was mounted on a petrographic microscope, as described in other
184 studies ([Esposito et al., 2016](#); [Kl ebeszt et al., 2015](#)). The heating procedure consisted in an
185 increase of the temperature of a crystal (with one or more MI) from room temperature up to
186 the temperature for melting crystals inside the MI to reverse the crystallization that occurred
187 after MI entrapment ([Narv ez et al., 2023](#)) (**Table 2**; **Supplementary Fig. 3**; **Supplementary**
188 **Note 1**). The experiments lasted from 2 to 3 hours, with relatively fast heating rates and a
189 limited duration for each heating plateau (**Table 2**). The cooling rate after quenching the
190 experiment is of ~ 190 °C/min. After the experiments, MI of interest were exposed on the
191 surface of crystals.

192 **3.4. Raman analyses**

193 Vapor bubbles in 30 bubble-bearing MI studied here were analyzed for volatile contents using
194 Raman spectroscopy with a Horiba Jobin Yvon LabRAM HR Evolution Raman System at the
195 Department of Earth and Environmental Sciences, University of Milano-Bicocca (Italy) (see
196 **Supplementary Note 1** for more details). CO₂ spectra were recorded using a green Nd 532.06
197 nm laser source, powered at 150 mW by mean of the 25% neutral density filter, with a 100 ×
198 magnification. Spectra have been acquired by one accumulation of 30 or 60 seconds each.
199 Spectra have been treated with baseline correction with a pseudo Voight function ([Yuan and](#)
200 [Mayanovic, 2017](#)) by the freeware software Fityk 1.3.1 (<https://fityk.nieto.pl/>) ([Wojdyr, 2010](#)).
201 The presence of CO₂ in bubbles in bubble-bearing MI is characterized by two bands at ~ 1285
202 cm^{-1} and ~ 1388 cm^{-1} , the Fermi doublet and two symmetrical weak bands below 1285 and
203 above 1388 cm^{-1} ([Colthup et al., 1975](#)). The distance between the Fermi doublet is proportional
204 to fluid density ([Frezzotti et al., 2012](#); [Garrabos et al., 1980](#)). Several calibrations have been
205 proposed for calculating the density of CO₂ fluid inclusions and in this study, we used the
206 densimeter of [Remigi et al. \(2021\)](#). The CO₂ content of the vapor bubbles can then be calculated
207 using the mass ratio between bubble and glass in the hosted MI ([Esposito et al., 2011](#); [Hartley](#)
208 [et al., 2014](#); [Moore et al., 2015](#)), with the following equation:

$$[CO_2]_{bubble} = \frac{M_{bubble}^{CO_2}}{M_{glass}} * 10^6 = \frac{V_{bubble} * CO_{2,density}}{V_{glass} * Melt_{density}} * 10^6 \#(1)$$

Where we use an estimated $Melt_{density}$ of 2.32 g.cm⁻³, using the Iqpet software (Carr and Gazel, 2017), V_{bubble} , which is the bubble volume, V_{glass} , the volume of the glass and $CO_{2,density}$ is the measured fluid density (Supplementary Table 1).

3.5. Analyses of volatiles by SIMS

Volatile (CO₂, H₂O, F, S, Cl) and Al concentrations of the glass in reheated MI were determined on an indium mount with the exposed MI of all crystals, using Secondary Ions Mass Spectrometry (SIMS), Cameca 1280 at Centre de Recherches Pétrographiques et Géochimiques (Nancy, France) (see Supplementary Note 1 for more details). These analyses were performed before EPM, to prevent a potential carbon contamination from carbon coating. Indium mounts were dried in a vacuum-oven for 12h at 40 °C and each sample was gold-coated. Cs⁺ was used as the primary ion source with the use of electron gun and a primary beam intensity of 0.5 nA. The contrast and field apertures were of 400 μm and 1200 μm, respectively and the mass resolving power was of 7007. The counting times on each mass of elements were of 4 s for ¹²C, ¹⁶O, ¹H, ³²S and ³⁵Cl, 2.96 s for ¹⁸O, ²⁷Al and ³⁰Si and 4.96 s for ¹⁹F. Each analytical spot was pre-sputtered for 300 s to clean the surface before analysis and the deadtime applied was of 35 ns. Detection limits are of 0.01 wt% for H₂O, 42 ppm for CO₂, 10 ppm for F, 2 ppm for S and 31 ppm for Cl. The analytical accuracy was calculated based on the average deviation of the individual reference glasses from calibration curves. This deviation was 1 % for CO₂, 2 % for H₂O, 2 % for F, 2 % S and 3 % for Cl. The high CO₂ contents measured in the MI of our study (> 3000 ppm) had a ¹²C⁺ signal that was stable during SIMS analysis, which is consistent with high intrinsic CO₂ contents rather than a contamination from surficial carbon.

3.6. Electron microprobe analysis of MI and host crystals

The compositions of the crystals and glasses were obtained using an electron microprobe (EMP; Jeol 8200 at the Earth Sciences Department, University of Milan), after SIMS analyses (see Supplementary Note 1 for more details). For residual glasses and MI, alkalis were analyzed first to minimize their migration under the electron beam. The EMP analytical conditions were set to an acceleration voltage of 15 kV, a beam current of 5 nA, a focused beam (spot size of ~1 μm) for minerals and a defocused 5-μm diameter beam size for glasses. For all elements, peak and background counting times, were set to 30 and 10 s, respectively. The oxides analyzed were Na₂O, K₂O, SiO₂, Al₂O₃, TiO₂, FeO, MgO, CaO, MnO, NiO, P₂O₅ and Cr₂O₃. The host mineral compositions were analyzed close to the MI. Typical analytical errors on major elements are 1% for CaO and K₂O, 2% relative for SiO₂, MgO and Al₂O₃, 3% for Na₂O, MnO, TiO₂, FeO, Cr₂O₃ and NiO and 4% for P₂O₅.

3.7. Temperature/pressure estimations

The equilibrium temperatures were estimated using the equations 33 for clinopyroxene-liquid thermometer and the equation 24a for plagioclase-liquid thermometer from Putirka (2008). The standard deviations of the estimates reported by Putirka (2008) are ± 41°C and 36°C, respectively. The fluid-saturation pressure of each MI was estimated using H₂O and CO₂

248 volatile contents and MagmaSat (Ghiorso and Gualda, 2015). These temperature and pressure
249 values were compared to the ones found based on the clinopyroxene-liquid thermometers and
250 barometers for alkaline differentiated magmas from Masotta et al. (2013).

251

252 4. Results

253 4.1. Samples and petrography of the MI

254 The analyzed trachyandesites present a 55.8-56.5 wt% SiO₂ content and a 7.4-9.1 wt%
255 Na₂O+K₂O content (Fig. 2; Table 3). Samples from Tabriz have between 10 and 24 vol% of
256 phenocrysts. These crystals are mainly plagioclase crystals up to 1 mm in size (6-7 vol%),
257 pyroxene up to 500 μm (0.5-2 vol%), few oxides (0.1-1 vol%) and altered minerals such as
258 altered amphibole and co-existing biotite (2-14 vol %). The Alborz sample has ~62 vol% of
259 crystals with millimetric to centimetric plagioclase (58 vol%), altered amphiboles and co-
260 existing biotite (3 vol%), clinopyroxene up to 700 μm (< 1 vol%), as well as few oxides (0.5
261 vol%) (Supplementary Note 1).

262 Melt inclusions are partially or highly crystallized in plagioclase crystals from the
263 trachyandesites, being completely opaque while they are homogeneous or containing small
264 crystals in clinopyroxene crystals (Supplementary Fig. 3). Petrographic descriptions of the
265 MI found in plagioclase and clinopyroxene, and other information on individual MIs and their
266 host crystals are listed in Supplementary Table 1 and in the Supplementary Note 2. MI are
267 mostly ellipsoidal in shape. Sizes and volumes of the MI have been estimated, as well as bubble
268 sizes and volumes for bubble-bearing MI (Supplementary Fig. 4). Most of the MI in
269 clinopyroxene and plagioclase crystals are between 3 and 22 μm in length.

270 For Alborz, 30 % of the MI that are hosted in heated plagioclase are between 3 and 22 μm,
271 while the rest of the MI in plagioclase are between 22 and 140 μm (Supplementary Fig. 4).
272 These MI reach the highest length, as 10% of MI in plagioclase from Alborz vary between 98
273 and 140 μm (Supplementary Fig. 4). The same relationship is observed for MI in
274 clinopyroxene for Alborz, where 50 % of them are between 3 and 22 μm in length and the rest
275 of the MI are up to 98 μm in length (with a lower percentage). For Alborz, 86 % of the MI in
276 heated plagioclase are bubble-bearing, while 14 % are glassy. For clinopyroxene from Alborz,
277 only 30 % of the MI are bubble-bearing (Supplementary Table 1). The distribution of the
278 bubble volume fractions for Alborz mostly have a peak at 1 vol% (35 and 37 % for
279 clinopyroxene and plagioclase, respectively), which is lower than the threshold corresponding
280 to shrinkage bubbles formed by post-entrapment modifications (< 5 vol. %) (Lowenstern, 2003,
281 1995) (Supplementary Fig. 4; Supplementary Note 2).

282 For Tabriz, 60 % of the MI in heated plagioclase are measuring 3 to 22 μm, while 30 % are
283 between 22 and 41 μm and the rest is up to 79 μm (Supplementary Fig. 4). For clinopyroxene,
284 50 % of the MI are 3 to 22 μm in length while the rest are measuring up to 60 μm
285 (Supplementary Fig. 4). For Tabriz, 70 % of the MI in heated plagioclase are bubble-bearing
286 while all clinopyroxene-hosted MI are glassy. For Tabriz, the bubble volume fractions of 59 %
287 of the heated plagioclase-hosted MI are superior to this threshold, going from 6 to 10 vol%,
288 with two outliers at 24 and 26 vol% (Supplementary Fig. 4).

4.2. Heating experiments

290 Examples of MI before, during and after the heating experiments are shown in **Fig. 3** and
291 **Supplementary Fig. 3**. Heating ramps for the plagioclase and clinopyroxene crystals are
292 reported in **Table 2**. During the heating ramps, melting started around 1000-1060 °C for
293 plagioclase-hosted MI from Alborz, while it began around 1090-1100 °C for the Tabriz ones
294 (**Supplementary Table 1**). All plagioclase-hosted MI were homogenized at 1120 °C
295 (**Supplementary Table 1**). During the experiments in clinopyroxenes, the visibility was not as
296 good as in plagioclase crystals so the exact temperature of beginning of melting is generally
297 not known. However, the MI were homogenized after being heated up to 1120 or 1160 °C for
298 Alborz and Tabriz, respectively. In most cases, plagioclase-hosted MI showed a homogeneous
299 silicate liquid plus a vapor bubble at the quenching temperatures (**Fig. 3a-g**). However, the
300 bubbles did not dissolve in the MI, even at high temperature after melting of the daughter
301 phases. For some clinopyroxene-hosted MI, the crystals did not dissolve into the melt with
302 further heating, suggesting that these MI represent heterogeneous trapping of melt and
303 minerals, and hence were not analyzed. These types of MI are classified in the literature as
304 mixed inclusions (Roedder, 1979). 47 % of the observed MI showed a homogenous liquid
305 phase at the temperature from which the MI was quenched (**Fig. 3h-k**). These MI that quenched
306 to a homogeneous glass phase show a maximum size of 60 μm (**Fig. 3h-k; Supplementary**
307 **Table 1**).

308

4.3. Major element compositions of MI

309 A total of 145 MI were brought to the surface of the crystals and the compositions of 56 of
310 them were quantified using EMP (**Supplementary Tables 1-2**). The others MI were either too
311 small, or fractured, or there was no room for analysis to avoid previous SIMS analyzed spots,
312 or the crystals were lost during the sample preparation at the different steps.

313 For the Alborz region, the eleven unheated MI hosted in clinopyroxene show a trachytic
314 composition, with the exception of one with a dacitic composition. The SiO_2 contents of these
315 unheated MI vary from 63.5 to 66.1 wt% and their $\text{Na}_2\text{O} + \text{K}_2\text{O}$ contents between 11.2-12.2
316 wt% (**Fig. 2**). The eight heated MI in clinopyroxene also show trachytic compositions, apart
317 from two MI presenting a dacitic affinity. In particular, SiO_2 varies from 61.4-65.8 wt% and
318 $\text{Na}_2\text{O} + \text{K}_2\text{O}$ from 5.7-11.7 for heated MI (**Fig. 2**). For the 16 heated MI hosted in plagioclase,
319 the MI compositions plot in the trachyandesitic field of the TAS diagram (**Fig. 2; Table 3**) (Le
320 Bas et al., 1986). The SiO_2 contents of these heated MI vary from 54.6 to 59.2 wt% and their
321 $\text{Na}_2\text{O} + \text{K}_2\text{O}$ contents between 8.0-9.8 wt% (**Fig. 2**). Three heated MI hosted in plagioclase lie
322 on the boundary with the trachytic field (60.3-62.1 wt% SiO_2 ; 9-10 wt% $\text{Na}_2\text{O} + \text{K}_2\text{O}$) (**Fig.**
323 **2**). Most of the MI in heated plagioclase of both samples are in the same field as the whole
324 rocks of the trachyandesites (**Fig. 2; Table 1**).

325 For Tabriz, the two unheated MI hosted in clinopyroxene show a slightly higher composition
326 than the Alborz ones (67.6-67.8 wt% SiO_2 ; 12.2-12.4 wt% $\text{Na}_2\text{O} + \text{K}_2\text{O}$). One heated MI hosted
327 in clinopyroxene shows a trachyandesitic composition (**Fig. 2**). The 18 heated MI hosted in
328 plagioclase from Tabriz are similar in composition to those from Alborz described above. In
329 fact, these MI show compositions between the trachyandesitic and trachytic fields (55.5-65.2
330 wt% SiO_2 ; 6.0-10.9 wt% $\text{Na}_2\text{O} + \text{K}_2\text{O}$) (**Fig. 2**).

331 In the Harker diagrams, all MI pertain to shoshonitic series, apart from three dacitic MI
332 (including an unheated clinopyroxene-hosted MI and two heated ones, poor in K₂O) (**Fig. 4a**).
333 One of the heated dacitic MI is likely to have been overheated, as it shows lower Na₂O and
334 K₂O and Al₂O₃ concentrations and high FeO, MgO and CaO concentrations. However, the two
335 other dacitic MI do not show anomalous compositions, so these MI could correspond to other
336 melts trapped by the crystals. The Al₂O₃, CaO and FeO contents of the trachyandesitic MI are
337 variable depending on the hosts (**Fig. 4b-d**). Heated MI in plagioclase from both regions have
338 higher CaO, FeO and MgO contents compared to MI in clinopyroxene crystals (**Fig. 4c-d**).
339 Whole rocks of the samples plot in the same field as the MI that have low SiO₂ contents (< 57
340 wt%) (**Fig. 4**), as in **Fig. 2**. For a given MgO value, SiO₂ of the whole rocks are lower than MI
341 compositions (**Fig. 4e**), while K₂O of the whole rocks are higher than some compositions of
342 heated MI in plagioclase but lower than the other clinopyroxene- and plagioclase-hosted MI
343 compositions (**Fig. 4f**).

344 4.4. Compositions of host minerals

345 MI hosts crystals are plagioclase that are mainly bytownite while few of them are labradorite
346 or anorthite, with Anorthite ($An = \frac{Ca}{Ca + Na + K}$) content of 73 ± 2 % for Alborz and slightly
347 higher for Tabriz (78 ± 11 %). Clinopyroxene are augites, with a similar Enstatite content (En
348 $= \frac{Mg}{Mg + Fe + Ca}$) of 42 ± 1 % or 41 ± 3 % for Alborz and Tabriz, respectively (**Supplementary**
349 **Fig. 5; Supplementary Table 2**). Plagioclase crystals sometimes show some zonations while
350 clinopyroxenes are mostly reverse-zoned, indicating changes in the magmatic storage
351 conditions ([Costa et al., 2020](#); [Costa and Morgan, 2010](#); [Petrone et al., 2016](#)). The Mg number
352 ($Mg\# = \frac{Mg}{Mg + Fe_{tot}}$) of the clinopyroxene host crystals from Alborz is nearly constant, from 0.67-
353 0.70, indicating that MI were trapped at the same degree of magma differentiation
354 (**Supplementary Fig. 6**). For Tabriz, the Mg# of two clinopyroxenes hosts is lower and is of
355 0.6 (**Supplementary Fig. 6**). For plagioclase, the anorthite content is more variable, with An
356 values mainly between 70 and 74 % (**Supplementary Fig. 7**). Some MI hosted in plagioclase
357 crystals show a high MgO and FeO content, in the An-rich host plagioclase (3-4 wt% in MgO,
358 10-13 wt% in FeO, for An₈₂₋₉₂) (**Supplementary Fig. 7**).

359 4.5. Raman in bubble-bearing MI

360 As CO₂ is poorly affected by diffusion through the host mineral but much more by the
361 formation of shrinkage bubbles ([Metrich and Wallace, 2008](#); [Portnyagin et al., 2008](#)), the
362 quantification of the amount of CO₂ in our bubble-bearing MI was carried out. 30 bubble-
363 bearing MI analyzed by Raman showed characteristic peaks of CO₂ (**Methods;**
364 **Supplementary Fig. 8**). However, signals were difficult to fit, because they either showed low
365 intensity peaks with less than 3 points above the background or showed a peak asymmetry.
366 Only one Raman signal in a bubble was successfully fitted, leading to a low CO₂ density of
367 0.05 g.cm⁻³ (**Supplementary Fig. 8; Supplementary Table 1**) likely affected by a large error
368 bar ([Remigi et al., 2021](#)). Using equation (1), and knowing the melt and bubble volumes [using
369 [Tucker et al. \(2019\)](#)], the CO₂ content in this bubble is 198 ± 40 ppm (with a worst-case
370 scenario error of 20%), representing 12% of the total CO₂ in the MI (see next section)
371 (**Supplementary Fig. 8; Supplementary Table 1**). This bubble comprised 1 % of the host MI
372 total volume (**Supplementary Fig. 8; Supplementary Table 1**).

4.6. Volatile content of the magma

374 35 MI were successfully analyzed for volatiles. For Alborz, 11 unheated MI in clinopyroxene,
375 4 heated MI hosted in clinopyroxene, and other 8 heated MI hosted in plagioclase were
376 analyzed by SIMS. For Tabriz, one heated MI hosted in clinopyroxene and 11 heated MI hosted
377 in plagioclase were analyzed by SIMS, to obtain a database of volatile contents of MI from
378 these two Iranian localities (**Supplementary Tables 1-2**).

379 Volatile contents detected in MI in clinopyroxene and plagioclase are shown in **Fig. 5-6** and
380 **Supplementary Fig. 9 (Supplementary Table 2)**. CO₂ contents are relatively high for the
381 degree of differentiation that the studied MI represent. In particular, up to 6733 ppm of CO₂
382 was detected in a bubble-bearing MI hosted in plagioclase from Tabriz, with a bubble
383 occupying 6 vol% of the MI (**Fig. 5a; Supplementary Table 2**). For unheated clinopyroxene-
384 hosted MI from Alborz, CO₂ contents range from 326 (± 3) ppm to 4243 (± 42) ppm, with an
385 average of 1827 ppm. In heated clinopyroxene crystals, CO₂ contents of MI range from 183 (\pm
386 2) ppm to 5272 (± 53) ppm. The values vary from 401 (± 4) ppm to 5668 (± 57) ppm (with an
387 average of 2101 ppm) for heated plagioclase-hosted MI from Alborz, while for Tabriz, they
388 are from 76 (± 1) to 6733 (± 67) ppm. The only clinopyroxene-hosted MI from Tabriz gives a
389 value of 2703 ± 27 ppm.

390 Water contents in MI are variable for unheated clinopyroxene-hosted MI from Alborz, varying
391 between 0.2 to 6.2 (± 0.1) wt%. Two different populations can be identified: one between 0.2
392 to 2 (± 0.04) wt% and another with higher contents: from 5 (± 0.1) to 6 (± 0.1) wt% (**Fig. 5b;**
393 **Supplementary Table 2**). H₂O contents are very low for MI in heated crystals, as reported
394 below (**Fig. 5b; Supplementary Table 2**).

395 In heated clinopyroxene crystals, H₂O range from 0.02 to 0.13 wt%, as well as in the heated
396 plagioclase crystals from Alborz. In heated plagioclase crystals from Tabriz, H₂O contents are
397 also significantly low, as 0.03 wt%, or under the detection limit.

398 By comparing the H₂O between unheated or heated crystals, some H may have been lost during
399 the experiments, even if caution was taken during the experiments to not stay too long at high
400 temperature (**Table 2**). However, as the H₂O data in heated clinopyroxene shows two different
401 populations (one with a relatively low content and another one with higher values;
402 **Supplementary Table 2**), no H₂O correction was applied to the heated minerals.

403 Fluorine contents in MI are higher in Alborz trachyandesite, compared to Tabriz ones (**Fig. 5c;**
404 **Supplementary Table 2**). F is variable for unheated clinopyroxene-hosted MI from Alborz,
405 between 167 (± 3) and 972 (± 19) ppm (with an average of 506 ppm). In heated clinopyroxene
406 crystals, concentrations of F range from 795 (± 16) to 1146 (± 23) ppm (with an average of 971
407 ppm). In heated plagioclase crystals, F is from 23 to 432 ppm (with an average of 238 ppm).
408 For Tabriz for MI in plagioclase, F is from 13 to 164 ppm. F contents of MI hosted in
409 clinopyroxene and some plagioclase crystals from Tabriz were under the detection limit of
410 SIMS.

411 Sulfur contents in MI are relatively low for clinopyroxenes but higher for plagioclase from
412 both the Alborz and the Tabriz areas (**Fig. 5d; Supplementary Table 2**). For unheated
413 clinopyroxene-hosted MI from Alborz, S is between 48 and 172 ppm (with an average of 89
414 ppm). For MI hosted in heated clinopyroxene crystals from Alborz, S detected values range
415 from 2 to 156 ppm (with an average of 83 ppm). In heated plagioclase crystals from Alborz,
416 the sulfur in MI is from 59 to 327 ppm, with a high value of 2347 ppm. For Tabriz in MI from
417 plagioclase, S is in the same range as for Alborz, from 14 to 2953 ppm (with an average of

418 1264 ppm). S content of one clinopyroxene-hosted MI from Tabriz was under the detection
419 limit.

420 For Chlorine, concentrations are higher in plagioclase-hosted heated MI from Alborz and
421 Tabriz than in clinopyroxenes (**Fig. 5e; Supplementary Table 2**). For unheated
422 clinopyroxene-hosted MI from Alborz, Cl is between 231 (± 7) and 852 (± 26) ppm (with an
423 average of 592 ppm). In heated clinopyroxene crystals, Cl detected values range from 550 (\pm
424 17) to 595 (± 18) ppm (with an average of 573 ppm). In heated plagioclase crystals, Cl contents
425 are from 49 to 1334 ppm (with an average of 831 ppm). For Tabriz heated plagioclase MI, Cl
426 contents are from 36 to 2953 ppm (with an average of 1264 ppm). Cl was below detection limit
427 in the clinopyroxene-hosted MI from Tabriz.

428
429 Halogen ratios, such as Cl/F, can be specific tracers of the origin of magmas, magma storage
430 conditions or eruption dynamic ([Balcone-Boissard et al., 2023](#)) (**Supplementary Fig. 9;**
431 **Supplementary Table 1**). A linear correlation is identified for MI in heated plagioclase and
432 unheated clinopyroxene crystals from Alborz (**Supplementary Fig. 9; Supplementary Table**
433 **1**), with Cl/F ratios between 1 and 3 for clinopyroxene or 1 and 4 for heated plagioclase, with
434 two higher values of 12 and 48. Such correlation indicates a common magma source. For
435 Tabriz, the data is too scarce to be able to conclude on the source but they show two high ratios
436 of 59 and 175. For Cl versus CO₂, Cl values seem to be constant while CO₂ varies (**Fig. 6b**).
437 No correlation seems to have been identified for S or F versus CO₂ (**Fig. 6c,d**). Cl versus F
438 shows a Cl-enrichment, as in other volcanic arcs (**Supplementary Fig. 9**) ([Rose-Koga et al.,](#)
439 [2014](#)).

440 Volatiles and K₂O contents are variable for plagioclase and clinopyroxene MI (**Supplementary**
441 **Fig. 10**). TiO₂ contents are constant for all volatiles for unheated clinopyroxenes
442 (**Supplementary Fig. 10**). For plagioclase, TiO₂ contents are variable depending on the
443 volatiles (**Supplementary Fig. 10**). SiO₂ contents for plagioclase MI are in the same ranges,
444 mainly from 54 to 59 wt% SiO₂, while volatile contents are spread out. SiO₂ values for unheated
445 clinopyroxene are also in the same range, from 64 to 68 wt% SiO₂ but their volatile contents
446 are dispersed (**Supplementary Fig. 10**).

447 **4.7. Estimates of magma storage temperatures and pressures**

448 For silicic melts, volatile saturation in melts is strongly controlled by pressure and significantly
449 less so by temperature and melt composition ([Wallace et al., 1999](#)), which allows to retrieve
450 pressures and depths of magmas based on MI volatiles concentrations.

451 The pre-eruptive magma storage temperatures using the clinopyroxene-liquid thermometer for
452 Alborz give lower temperatures for MI in unheated crystals compared to heated clinopyroxene
453 (859 ± 39 °C and 988 ± 58 °C, respectively) (**Supplementary Fig. 11; Supplementary Table**
454 **3**) ([Putirka, 2008](#)). No temperature was calculated for MI in heated cpx of Tabriz (lack of data).
455 For plagioclase-hosted heated MI, the estimated temperatures were calculated using the
456 plagioclase-liquid thermometer ([Putirka, 2008](#)) and are of the same order of magnitude for
457 Alborz and Tabriz (1083 ± 11 °C and 1087 ± 24 °C, respectively) (**Supplementary Fig. 11;**
458 **Supplementary Table 3**). Temperatures calculated for MI of Alborz using the clinopyroxene-
459 liquid geothermobarometer of [Masotta et al. \(2013\)](#) are higher compared to the ones calculated
460 with [Putirka \(2008\)](#) (**Supplementary Fig. 11, Supplementary Table 4**). In fact, the
461 temperatures estimates for Alborz are 1010 ± 22 °C for unheated MI in clinopyroxene and 1022
462 ± 47 °C for heated MI in clinopyroxene (**Supplementary Fig. 11, Supplementary Table 4**).

463 Most of the studies are focused on the reconstruction of H₂O and CO₂ because they allow to
464 calculate the pressure of entrapment of MI if we assume that the magma was at the H₂O-CO₂
465 saturation (Newman and Lowenstern, 2002). For both Alborz and Tabriz, nine pressures were
466 calculated for MI using MagmaSat calculations (Ghiorso and Gualda, 2015). These pressures
467 are mostly ranging from 60 to 310 MPa for all crystals. Few outliers either show very high
468 pressures, with three pressures higher than 900 MPa (Supplementary Fig. 11, Supplementary
469 Table 3). Assuming a magma density of 2.3 g.cm⁻³ (Carr and Gazel, 2017), the pressure range
470 given by the MI (60-310 MPa) correspond to 3-14 km in depth. Nevertheless, the MI having
471 H₂O-CO₂ contents indicating high storage pressures of 956, 1065 MPa and even 1451 MPa
472 correspond to entrapment depths of 42-64 km (Supplementary Fig. 11, Supplementary
473 Table 4).

474 Using the clinopyroxene-liquid geothermobarometer of Masotta et al. (2013), calculated
475 pressures show a good overlap relative to those calculated based on MagmaSat. For instance,
476 for unheated clinopyroxene MI from Alborz, pressures based on clinopyroxene and MI
477 compositions are 150 ± 65 MPa, while they are 194 ± 55 MPa for heated clinopyroxene MI
478 from Alborz (Supplementary Fig. 11, Supplementary Table 4). The corresponding depths
479 of these ranges of pressures (93-297 MPa) range from 4 to 13 km depth, assuming a magma
480 density of 2.3 g.cm⁻³ (Carr and Gazel, 2017). These depths are in the same order of magnitude
481 of pressure of the main range determined by MagmaSat.
482

483 5. Discussion

484 5.1. Petrological implications of the volatile analyses

485 The volatile measurements in the MI of the Eocene Iranian volcanics allow us to estimate the
486 pre-eruptive CO₂ content of these magmas. As some MI are bubble-bearing, the
487 representativity of the CO₂ analyses in the glass needs to be assessed. The MI compositions
488 can be modified by post-entrapment processes and re-equilibration of fast-diffusing elements
489 such as H if the magma was stored for an extended period of time (Bucholz et al., 2013; Gaetani
490 et al., 2012; Portnyagin et al., 2008; Wallace et al., 2021). Regarding C, the CO₂ content can
491 be re-distributed in bubbles in the MI, if bubbles are formed during post-entrapment
492 crystallization (Steele-Macinnis et al., 2011) or as shrinkage bubbles in response to cooling
493 (Lowenstern, 1995; Metrich and Wallace, 2008; Roedder, 1979). Some studies assess the CO₂
494 concentrations by summation of the CO₂ contents of the glass determined by SIMS and the
495 CO₂ contained in the bubble(s) (Esposito et al., 2011; Hartley et al., 2014; Metrich and Wallace,
496 2008; Tucker et al., 2019). According to our Raman analyses, bubbles in bubble-bearing MI
497 showed the presence of CO₂ but with very low densities (Supplementary Fig. 8;
498 Supplementary Table 1). Considering only the MI where the CO₂ in the bubble was
499 satisfyingly quantified, we found ~198 ppm of CO₂ in the bubble (Supplementary Fig. 8;
500 Supplementary Table 1). As the bubble represents ~1 % of the total volume of the MI and as
501 the CO₂ concentration in the glass was 1438 ppm, the bubble may contain ~12 % of the total
502 CO₂ in the inclusion (Supplementary Table 1). This percentage is low considering that
503 bubbles can contain up to 90 % of the total CO₂ of MI (Esposito et al., 2011; Hartley et al.,
504 2014; Moore et al., 2015). Therefore, the combination of weak Raman CO₂ signals, low bubble
505 CO₂ density, and generally low bubble volume percentages, suggests that MI need limited
506 correction for CO₂ content lost in the bubble. Assuming that these measurements are
507 representative of our bubble-bearing MI set, MI CO₂ contents could be corrected by adding 12

508 % to the measured values. However, in this study, we dismissed this correction as the potential
509 12 % value is based on only one MI out of 35. SIMS-MI CO₂ values were thus not corrected
510 to avoid introducing bias and produce conservative, but reliable, measurements, thus
511 representing minimum values.

512 The volatile content further provides important information on melt storage pressures,
513 fostering our understanding of the plumbing system and degassing behavior of the Iranian
514 magmas. Considering the barometer using H₂O and CO₂ contents of the MI (MagmaSat) or
515 clinopyroxene compositions and H₂O contents (Masotta et al., 2013), magmas were stored at a
516 wide range of pressures, mainly corresponding to depths between 3 and 14 km. A trans-crustal
517 magmatic plumbing system with various storage depths is involved, in accordance with the
518 wide range of pressures obtained (60-310 MPa). These pressure estimates can be affected by
519 secondary processes such as H₂O loss (or gain) due to diffusion at high temperature after MI
520 formation. This can occur either before eruption during shallow crustal melt storage, or during
521 deposition after eruption, or during high-temperature heating experiments (Bucholz et al.,
522 2013; Esposito et al., 2023; Gaetani et al., 2012; Portnyagin et al., 2008; Wallace et al., 2021).
523 While the former two processes are difficult to estimate for MI hosted in plagioclase and
524 clinopyroxene crystals, the latter process was mitigated by short experiment duration from 2 to
525 3 hours and relatively short time spent at high temperatures, between 15-20 min above 1100
526 °C (Table 2). The combination of low H₂O and high CO₂ concentrations and, thus, low

527 $\frac{H_2O}{H_2O + CO_2}$ ratios in silicate melts may indicate an external income of CO₂ from deeper levels
528 within the magma plumbing system, driving melt dehydration and crystallization at shallower
529 levels (CO₂ flushing) (Caricchi et al., 2018). However, K₂O MI concentrations are higher for
530 low An# plagioclase, and are uncorrelated to CO₂ concentrations, whereas a correlation
531 between K₂O content of the MI and CO₂ is expected if CO₂ flushing occurs (Supplementary
532 Fig. 7a; 10b1) (Caricchi et al., 2018). Thus, our MI may not be representative of CO₂ flushing.
533 As previously discussed (Fig. 6a), the lack of correlation between CO₂ in MI with An# or Mg#
534 of the host minerals for Alborz or Tabriz MI (Supplementary Fig. 12) may further indicate
535 that water loss should be ascribed to other primary/secondary processes, the recognition of
536 which is beyond the scope of this study. Thus, our pressure estimates derived from H₂O and
537 CO₂ contents are considered here as minimum values. Independent of the accuracy of the
538 magma crystallization depths, the total CO₂ concentration of studied MI is not affected by the
539 H-loss by diffusion, with the assumption that the MI were originally trapped as a single
540 homogeneous silicate liquid phases.

541 Plagioclase crystals in both samples crystallize at a relatively earlier stage of differentiation
542 compared to clinopyroxene crystals, as shown on the SiO₂ vs MgO diagram, as SiO₂
543 concentrations are lower for relatively higher MgO contents (Fig. 4e). In general, MI and their
544 hosts composition show a reasonably good correlation compatible with simple crystal
545 fractionation (Supplementary Fig. 6-7). Considering these two points, plagioclase crystals
546 may have crystallized earlier than clinopyroxene. Furthermore, Rhyolite-MELTS runs carried
547 out using a trachyandesitic initial melt composition from our MI set (PS2-FD1-2-Z8-MI8)
548 allow to model the liquid line of descent of the system, with isobaric conditions (310 MPa;
549 pressure of the MI of interest), varying T (Fig. 2) and without considering CO₂ (Ghiorso and
550 Gualda, 2015; Gualda et al., 2012) (Supplementary Table 5). Oxygen fugacity has been either
551 set without a buffer or the NNO or QFM buffers but the oxidation state does not appear to
552 significantly influence the results. The modelled liquid lines of descent show an evolution

553 consistent with the whole-rocks and the MI compositions hosted both in plagioclase and
554 clinopyroxene (**Fig. 2**). A kink in the curves is observed in trachytic compositions, where Na
555 increases in plagioclase crystals. Clinopyroxene compositions change from En₅₈₋₆₀ to En₃₈₋₄₁ at
556 a temperature around ~1050 °C, and are comparable to the ones investigated here. The liquid
557 then evolves to rhyolitic composition with decreasing temperature (up to ~880 °C). The
558 predicted order of appearance of mineral phases is in good agreement with the MI compositions
559 and the compositions of the hosts, especially for clinopyroxene (**Fig. 2**). These Rhyolite-
560 MELTS runs are also in favor of the crystallization of plagioclase before the low-En
561 clinopyroxene. Plagioclase crystals could thus have trapped a trachyandesitic liquid, while
562 fractional crystallization led to clinopyroxene trapping trachytic, more evolved melts.

563

564 5.2. CO₂ budget of ascending melts and climatic implications

565 5.2.1. Assessment of the primary CO₂ budget

566 The CO₂ measured in MI from Alborz and Tabriz is similar for heated and unheated
567 crystals (**Fig. 5**; overlapping errors bars in **Fig. 7**; **Supplementary Fig. 13**; **Supplementary**
568 **Table 1**). Therefore, the CO₂ budget of ascending melts recorded in plagioclase- and
569 clinopyroxene-hosted MI can be assessed by considering all CO₂ values as a unique range, with
570 the minimum, maximum and average values of all CO₂ contents in these MI that are 76, 6733,
571 and 1757 ppm, respectively (**Fig. 7**; **Table 4**).

572 It is important to keep in mind, however, that the maximum CO₂ value of MI using
573 SIMS is a minimum estimate of the pre-eruptive degassed CO₂ of ascending melts since
574 additional CO₂ may have been lost during the ascending path of the magma from the source to
575 the lower crust. The volume of middle-late Eocene (~45-36 Ma) Iranian volcanics were
576 quantified between 7.6×10⁴ and 3.8×10⁵ km³ (van der Boon et al., 2021), based on a
577 compilation of available geological maps (Sahandi et al., 2014) and assuming thicknesses
578 between 2 and 10 km of volcanic formations from the literature (Berberian and King, 1981;
579 Morley et al., 2009; Stöcklin, 1974; van der Boon et al., 2021, and references therein; Verdel
580 et al., 2011). To the best of our knowledge, the Central Eocene volcanics that are buried beneath
581 a few-km-thick sequence of Oligo-Miocene sediments were not included in these calculations,
582 as only outcropping volcanics were considered. The volumes estimates mentioned above
583 should thus be considered as minimum values. To estimate the outflux of CO₂, f_{CO_2} , from the
584 SIMS measurements, the following equation was used (2):

$$585 f_{CO_2} = \frac{C_{CO_2, SIMS} * m_T}{t} \#(2)$$

586 where $C_{CO_2, SIMS}$ is the CO₂ content of MI measured with SIMS, m_T is the total mass of parental
587 magma in tons (t) and t is the duration of the eruptive event in years.

588 Assuming an average density of the magma of 2.32 g cm⁻³ (using the Igpert software; Carr and
589 Gazel, 2017) and referring to the magma volumes estimated by van der Boon et al. (2021), the
590 minimum and maximum mass erupted during the middle Eocene are 2-9×10¹⁴ t (**Table 4**).
591 Considering 70 % of magma crystallization to form a trachyte from a basalt (Macdonald et al.,
592 2008), the range of total mass of parental magma would be 3×10¹⁴-1.5×10¹⁵ t. The main source

593 of uncertainty in the CO₂ outflux comes from the magma volume estimates. We overcome this
594 limitation by accounting for the minimum and maximum values of the magma volume
595 estimated from [van der Boon et al. \(2021\)](#). (1) Accounting for the estimated pre-eruptive CO₂
596 budget of magmas as derived from the CO₂ concentrations of MI (76-6733 ppm) and the
597 minimum estimate of the total mass of Iran parental magma from 45-36 Ma (3×10^{14} t), the
598 outflux of CO₂ during this period is of 0.003-0.22 Mt CO₂/yr (**Table 4**). This CO₂ flux
599 corresponds to a total CO₂ emission of 2.28×10^{16} - 2.02×10^{18} g CO₂ (**Table 4**). Converting our
600 CO₂ flux in carbon (C), the C flux would be of ~0.001-0.061 Mt C/yr, corresponding to a total
601 C emission of 6.22×10^{15} - 5.51×10^{17} g C during the middle-late Eocene (~6-551 Pg C) (**Fig. 8**;
602 **Table 4; Supplementary Table 1**). (2) Accounting for the estimated pre-eruptive CO₂ budget
603 of magmas as derived from the CO₂ concentrations of MI (76-6733 ppm) and the maximum
604 estimate of the total mass of Iran parental magma from 45-36 Ma (1.5×10^{15} t), the outflux of
605 CO₂ during this period is of 0.01-1.12 Mt CO₂/yr (**Table 4**). This CO₂ flux corresponds to a
606 total CO₂ emission of 1.14×10^{17} - 1.01×10^{19} g CO₂ (**Table 4**). Converting our CO₂ flux in carbon
607 (C), the C flux would be of 0.003-0.306 Mt C/yr, corresponding to a total C emission of
608 3.11×10^{16} - 2.75×10^{18} g C during the middle-late Eocene (~31-2800 Pg C) (**Fig. 8; Table 4**;
609 **Supplementary Table 1**). The relative uncertainty on the outflux of CO₂ is 4 %, based on the
610 propagation of measurement uncertainty of CO₂ by SIMS and the age uncertainties reported
611 by [van der Boon et al. \(2021\)](#). Since our CO₂ SIMS measurements are likely to underestimate
612 the true CO₂ content of Iranian magmas, we can consider estimates calculated with the highest
613 CO₂ measured concentration by SIMS (6733 ppm of CO₂): 0.061-0.306 Mt C/yr (551-2800 Pg
614 C). Following our pressure and temperature estimates, and CO₂ degassing budget, a schematic
615 crustal section of the Iran Alborz and Tabriz magmas is shown on **Fig. 8**.

616 5.2.2. Assessment of possible CO₂ degassing from intruded carbonate 617 rocks

618 The amount of C released during volcanic episodes increases substantially when
619 eruptions occur within carbonate-rich rocks due to contact metamorphism of carbonates, as
620 well as CO₂ released by magmatic assimilation of carbonates ([Aiuppa et al., 2017](#); [Deegan et al., 2010](#);
621 [Iacono-Marziano et al., 2009](#); [Lee et al., 2013](#); [Lee and Lackey, 2015](#)). However, the
622 fraction of CO₂ released by magmatic assimilation of carbonates is not constrained for Iran.
623 Furthermore, most of the MI studied here record depths which are deeper than those of the
624 Meso-Paleozoic sedimentary carbonate sequence below those two studied localities in Iran
625 (~1.5 km and ~9-11 km) ([Allen et al., 2003](#); [Geological Survey and Mineral Exploration of Iran, 2006, 1995](#);
626 [Madanipour et al., 2018](#)). Therefore, the CO₂ of most studied MI is likely to
627 be of magmatic origins. [Van der Boon et al. \(2021\)](#) and [Zhang et al. \(2023\)](#) used a range of
628 “limestone factors” to calculate the whole amount of CO₂ released during Eocene volcanic
629 episodes of Iran and Tibet, respectively. Limestone factors involving 3.6–8.6 times higher CO₂
630 release by volcanic outgassing alone were constrained based on estimates of the CO₂ released
631 from carbonate sediments during the emplacement of the Emeishan large igneous province
632 (China) in the end-Guadalupian ([Courtillot and Renne, 2003](#); [Ganino and Arndt, 2009](#); [Self et al., 2006](#)).
633 As these limestone factors were estimated based on a Large Igneous Province in
634 China, which does not necessarily correspond to our geological context, we preferred not to
635 use them to calculate the final C flux. However, it is important to keep in mind that Eocene
636 volcanism in Iran erupted in shallow marine basins and through carbonate-rich rocks
637 ([Berberian and King, 1981](#)). We therefore warn the reader that the CO₂ outgassing due to

638 Iranian magmatism could be up to about one order of magnitude larger than values provided
639 above (0.061-0.306 Mt C/yr).

640 5.2.3. Contribution of Iranian magmatism to the MECO

641 The MECO is accompanied by a rise in atmospheric CO₂ concentrations (Bijl et al.,
642 2010; Henehan et al., 2020; Judd et al., 2024; Ma et al., 2024), warming in surface and
643 intermediate deep-water, extensive deep-sea carbonate dissolution (Bohaty et al., 2009) and
644 marine biotic change (Bijl et al., 2010; Cramwinckel et al., 2019; Edgar et al., 2013; Witkowski
645 et al., 2012). Sluijs et al. (2013) and van der Ploeg (2018) showed, with reconstructions and
646 simulations of the carbon cycle, an imbalance in the long-term inorganic carbon cycle which
647 is likely to have been caused by enhanced volcanism and a decrease in continental silicate
648 weathering during the MECO, in contrast to expectations from the silicate weathering
649 thermostat hypothesis. Indeed, van der Ploeg et al. (2018) surmised, based on osmium isotope
650 records and carbon and osmium cycle modeling, that global early and middle Eocene warmth
651 gradually diminished the weatherability of continental rocks and the strength of the silicate
652 weathering feedback, leading to the prolonged accumulation of volcanic CO₂ in the oceans and
653 atmosphere during the MECO. Foraminifera boron isotope ratios and carbon cycle simulations
654 constrained by available proxy data led to estimates of the carbon cycle imbalance during the
655 MECO around 2×10^{18} g C to 4×10^{18} g C (2000-4000 Pg C) (Henehan et al., 2020; Sluijs et al.,
656 2013; van der Boon et al., 2021; van der Ploeg et al., 2018).

657 New ages of Iranian Eocene volcanics provided by van der Boon et al. (2021) and
658 previously published radiometric ages showed that many volcanic areas in Iran were active
659 simultaneously around 40 Ma (with a clustering of ages during the middle Eocene). The MECO
660 thus corresponds to a phase of intense volcanism in this area, even if the uncertainty in the
661 individual samples ages is larger than the duration of the MECO (van der Boon et al., 2021).
662 Van der Boon et al. (2021) have taken the Deccan traps as a term of comparison as CO₂
663 degassing rates for continental arcs may be similar to (Marty and Tolstikhin, 1998) or larger
664 than for continental flood basalts (McKenzie et al., 2016; Wignall et al., 2009). These authors
665 calculated the amount of CO₂ degassed from these magmas based on a relationship between
666 lava volume and CO₂ emitted during the emplacement of the Deccan Traps (Tobin et al., 2017),
667 as no constraints on CO₂ emissions from these Iranian magmas existed so far. From this
668 method, the minimum estimate for C released from middle Eocene volcanism in Iran ranges
669 between 292-1461 Pg C (van der Boon et al., 2021). These C estimates increase to 1052-12565
670 Pg C considering the possible “limestone factor” (Berberian and King, 1981; Ganino and
671 Arndt, 2009; van der Boon et al., 2021). Our C estimates derived from MI analyses (551-2800
672 Pg C) are within the range and even slightly higher than the total magmatic C estimate by van
673 der Boon et al. (2021) or in their low range of the total C outgassed considering carbonates.
674 Furthermore, our C total estimates are also in the range of the carbon anomaly that drives the
675 MECO in carbon cycle simulations constrained by available proxy data (2000-4000 Pg C)
676 (Henehan et al., 2020; Sluijs et al., 2013; van der Boon et al., 2021; van der Ploeg et al., 2018),
677 and consistent with enhanced volcanism from the Azerbaijan-Bazman arc contributing to the
678 MECO (Ma et al., 2024).

679 Plate boundary evolution and carbon fluxes models have been used to estimate a net C
680 outgassing of ~46 Mt C/yr at around 40 Ma (Wong et al., 2019), whereas proxy data (e.g. CO₂
681 from leaf stomata, pedogenic carbonate $\delta^{13}C$, boron isotopes in foraminifera) suggest

682 significantly lower values of ~ 4.5 Mt C/yr (Foster et al., 2017). Considering that our maximum
683 estimate of degassing flux from the Iranian magmas (0.3 Mt C/yr) could be one order of
684 magnitude higher due to the limestone contribution, Iranian flare up emissions could account
685 for ~ 6 -60 % of the estimated net C outgassing. Apart from the Iranian flare-up, other magmatic
686 events have occurred in the middle Eocene: (i) large volumes of magmas were erupted from
687 multiple volcanic centers in the Basin and Range (Gans et al., 1989), (ii) increased arc
688 volcanism around the Pacific rim, with the ignimbrite flare-up in the Sierra Madre Occidental
689 of Mexico, (iii) an episode of magmatism in the East African Rift zone, as well as (iv) mid-
690 ocean ridge volcanism in the North Atlantic (Ma et al., 2024; van der Ploeg et al., 2018, and
691 references therein). While these magmatic events and associated degassing likely affected the
692 overall atmospheric C budget, their timing and magnitude are presently not sufficiently well
693 resolved to be linked with the MECO (van der Ploeg et al., 2018). The upper limit of ~ 60 %
694 contribution due to C outgassing from Iranian magmas around 40 Ma seem relatively high.
695 However, considering the length of the Neo-Tethyan subduction along the Arabian margin with
696 respect to the global arc and mid oceanic ridges length in the middle Eocene, even a
697 contribution of $\sim 6\%$ is significant. We emphasize that our C emission estimates are
698 conservative, as CO_2 -saturated magmas could have contained more CO_2 and could have been
699 stored deeper than the storage depth of the targeted MI. We thus provide strong observational
700 evidence in support of a major contribution to the MECO from Iranian magmatic flare up and
701 degassing. We further stress that petrological quantifications of the volatile budget of magmas
702 from arc settings are critical to improve estimates of magmatic CO_2 emissions and assess the
703 possible climatic effects of volcanic degassing. Once sufficient data will be collected, these
704 CO_2 estimates could even be used to constrain climate and/or carbon outflux models and
705 improve understanding of past climate drivers and dynamics (Arndt et al., 2011;
706 Castrogiovanni et al., 2024; Donnadieu et al., 2006; Lefebvre et al., 2013).

707 5. Conclusions

708 We provide the first quantification of the volatile content of unheated and heated MI in
709 clinopyroxene and plagioclase from Eocene Iran trachyandesites. Our study shows that the MI
710 are CO_2 -rich and derive from trachyandesitic to trachytic melts. These data allow us to provide
711 the first observation-based estimates of the amount of CO_2 degassed during the regional
712 middle-late Eocene magmatic flareup, which range between 2.28×10^{16} - 1.01×10^{19} g CO_2 and
713 thus 6.22×10^{15} - 2.75×10^{18} g C. Taking the maximum CO_2 concentration from SIMS
714 measurements, the estimated total C outflux is 0.061-0.306 Mt C/yr and could be up to an order
715 of magnitude higher considering the likely limestone contribution from interactions between
716 magmas and carbonate-rich rocks. This degassed CO_2 from Iran is likely to have provided a
717 significant contribution to the Middle Eocene Climatic Optimum. Future MI-based estimates
718 of CO_2 degassing and better constraints on the magma volumes will enable us to improve the
719 quantification of the magmatic contributions of Neo-Tethyan magmas to the MECO and,
720 possibly, other long-term Cenozoic climate trends.

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745 **Author contributions**

746 PBa provided the Iran samples. Conceptualization: PS, PBo, RE; Data curation: LO; Formal
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748 LO, PBo, VAM; Resources, supervision and validation: PS, RE, PBo, PBa; Visualization: LO;
749 Writing of the original draft: LO; Reviews and editing: all authors.

750

751 **Statements and declarations**

752

753 Competing interests: All the authors declare no competing interests.

754 **Data Availability**

755

756 All data generated or analyzed during this study are included in this manuscript (and its
757 supplementary information files).

758

759 **Figure captions (colors should be used for all the figures in print)**

760 **Figure 1: Maps of Iran. (a) Digital elevation map of the study region** with sampling sites
761 in the regions of Tabriz and the Alborz mountains are shown, as well as the distribution of
762 Eocene volcanic rocks after previous maps (Agard et al., 2011; Allen and Armstrong, 2008;
763 Nalivkin, 1976; van der Boon et al., 2017); **(b) Topography of Iran showing the main Zagros**
764 reverse fault, which is interpreted to coincide with the Arabian-Iranian suture zone (Ballato et
765 al., 2013). In **(a-b)**, a global topography Digital Elevation Model was used. “UDMA” refers to
766 the Urumieh–Dokhtar Magmatic Arc and “Mtns” refers to Mountains.

767 **Figure 2: Total Alkali Silica (TAS) diagram (Le Bas et al., 1986) of unheated and heated**
768 **clinopyroxene and plagioclase-hosted melt inclusions (MI) from Alborz and Tabriz**
769 **(Iran), as well as liquid compositions predicted by Rhyolite-MELTS.** Whole-rocks of the
770 samples are also shown. Melt inclusions literature data from Iran are also shown for
771 comparison, such as the olivine-hosted MI of the Pliocene-Quaternary lavas from the Qorveh-
772 Bijar volcanic belt (Salehi et al., 2020). Whole rocks from Alborz basalts and phonotephrite
773 are also shown (Verdel et al., 2011), as well as the host rocks of the olivine-hosted MI from
774 Salehi et al. (2020). For Rhyolite-MELTS predictions (Ghiorso and Gualda, 2015; Gualda et
775 al., 2012), the composition of a trachyandesitic MI (PS2-FD1-2-Z8-MI8) was used with an
776 isobaric run (at 310 MPa, found for this specific MI), with a varying temperature (from the
777 liquidus temperature to ~880 °C) with different oxidizing conditions (no buffer, NNO and
778 QFM). The appearance and disappearance of orthopyroxene (“opx”), clinopyroxene (“cpx”),
779 plagioclase (“plag”) and K-feldspar at low temperature (sanidine, “san”) predicted by Rhyolite-
780 MELTS are highlighted.

781 **Figure 3: Photomicrographs of MI as found in plagioclase and clinopyroxene crystals**
782 **before heating experiments, during and after heating. a-g) Partially crystallized melt**
783 inclusion in a plagioclase, from ambient temperature **(a)** to complete homogenization **(f)** and
784 after quenching **(g)**. **h-k)** Melt inclusions in clinopyroxene crystals, before the experiment and
785 after quenching. No pictures from the experiments are shown because the visibility was low.

786 **Figure 4: Harker diagrams for unheated and heated clinopyroxene- and plagioclase-**
787 **hosted MI. a) K₂O vs SiO₂; b) Al₂O₃ vs SiO₂; c) CaO vs SiO₂ and d) FeO vs SiO₂.** Whole-
788 rocks are also shown.

789 **Figure 5: Volatile contents depending on the MI hosts in trachyandesites from Alborz or**
790 **Tabriz (unheated or heated clinopyroxene and plagioclase crystals). a) CO₂ contents in**
791 MI, depending on the host and sample; **b) H₂O; c) F; d) S; e) Cl.** The error bars on SIMS
792 measurements are lower than the symbols.

793 **Figure 6: Volatile content from SIMS of unheated and heated MI hosted in clinopyroxene**
794 **and plagioclase crystals from Alborz and Tabriz. a) CO₂ and H₂O contents in MI; b) Cl**
795 versus CO₂ concentrations in MI; **c) S versus CO₂ concentrations in MI; d) F versus C₂O**
796 concentrations in MI. Error bars correspond to SIMS uncertainties. The error bars on SIMS
797 measurements are lower than the symbols.

798 **Figure 7: Whisker plot of the CO₂ contents in unheated or heated clinopyroxene and**
799 **plagioclase crystals. The center line denotes the median value (50th percentile), while the**
800 **box contains the 25th to 75th percentiles of the dataset. The black whiskers mark the 5th**

801 and 95th percentiles, and values beyond these upper and lower bounds are considered outliers,
802 marked with empty dots.

803 **Figure 8: Simplified schematic crustal section of the Iran Alborz and Tabriz magmas and**
804 **CO₂ degassing flux estimates during the middle-late Eocene. Supposing volatile**
805 **saturation in Iran magmas considering the wide range of CO₂ concentrations, H₂O and**
806 **CO₂ contents were considered as representative of the Iran magmatic pressures,**
807 **temperatures and depths at Eocene times. The CO₂ flux range presented in this figure**
808 **was calculated with the maximum CO₂ concentration measured by SIMS in the MI (see**
809 **discussion; Table 4). The estimation of the thickness of the carbonate rocks (~1.5 km and**
810 **deeper strata of unknown accurate thickness and depth) (Allen et al., 2003; Geological**
811 **Survey and Mineral Exploration of Iran, 2006, 1995; Madanipour et al., 2018) as well as Moho**
812 **depth estimates are from present time in Tabriz and Alborz region. The Moho depth**
813 **estimate is from a comparison between gravimetric Moho solutions and existing regional**
814 **seismic studies: 50 ± 2 km (Ebadi et al., 2019). In the Eocene, the substratum had already**
815 **undergone several phases of deformation; thus, for simplicity, strata are depicted here as sub-**
816 **horizontal (Madanipour et al., 2018).**

817 **Tables captions**

818

819 **Table 1: Description of the samples from Iran studied here (location, coordinates (in the**
820 **World Geodetic System WGS84) and nature of the samples are specified).**

821 **Table 2: Heating ramps used for experiments on crystals from Iran. The durations of**
822 **maintaining the temperature constant after each temperature step are also specified.**

823 **Table 3: Whole rock major and trace elements of the Iranian samples from Alborz and**
824 **Tabriz. The sampling locations are specified in Table 1. Analyses have been performed**
825 **at Bureau Veritas Mineral Laboratories (Canada) by ICP-OES (Inductively Coupled**
826 **Plasma-Optical Emission Spectrometry). The type of deposit is specified. Data are**
827 **expressed with total Fe as Fe³⁺ with LOI detailed.**

828 **Table 4: Magmatic CO₂ flux mass balance calculations for Iran magmas assuming CO₂**
829 **concentrations of MI by SIMS. CO₂ values could be higher considering CO₂ released**
830 **from carbonate sediments.**

831

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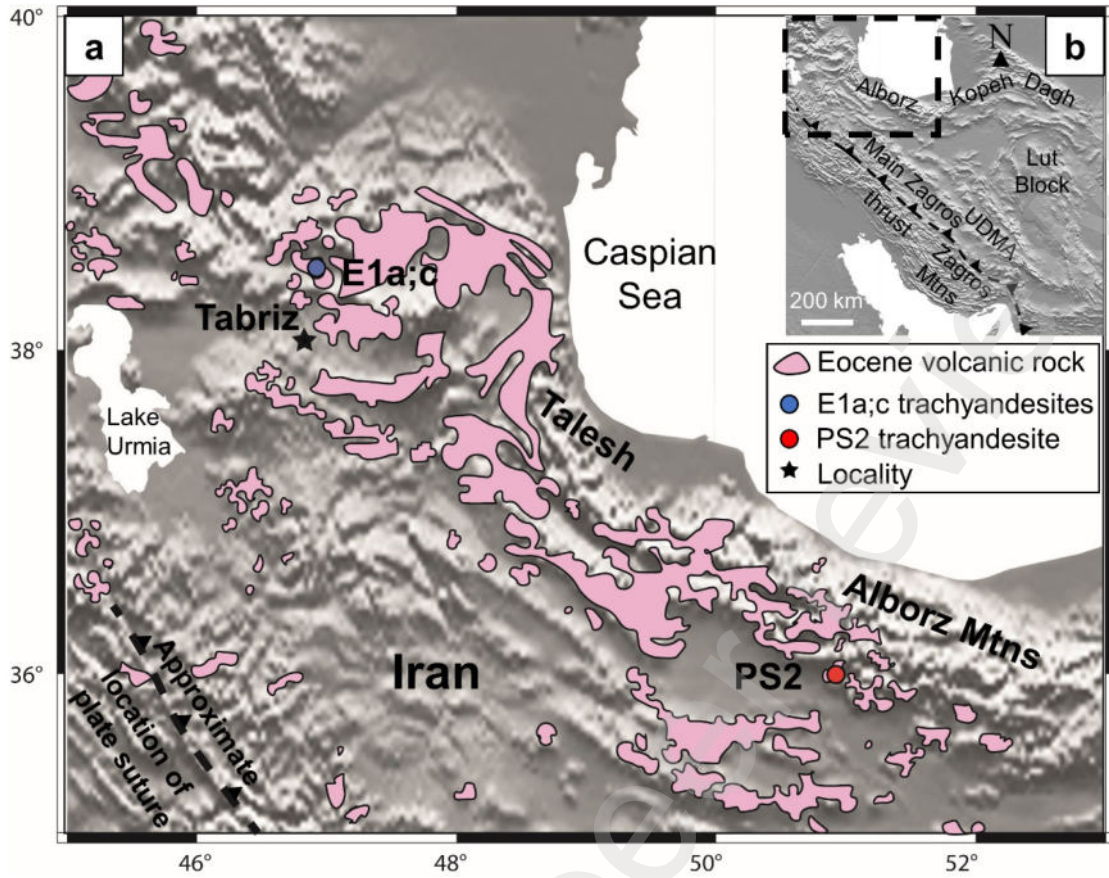
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1230 **Figures**

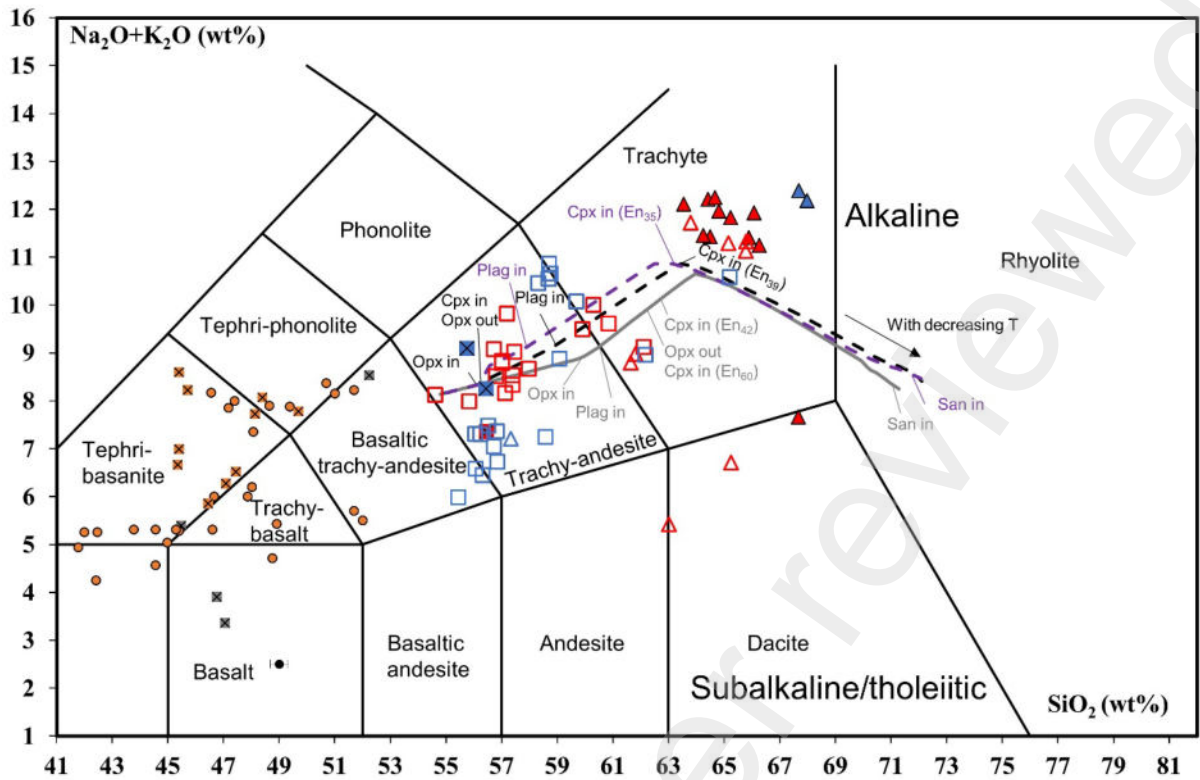
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1234 **Figure 1**



Iran Eocene

- Alborz
- Tabriz

MI hosts

- ▲ Clinopyroxene
- △ Heated clinopyroxene
- Heated feldspar
- ⊠ Whole rock

Rhyolite-MELTS simulations

- Basaltic trachy-andesite MI – no buffer
- - Basaltic trachy-andesite MI – NNO
- · - Basaltic trachy-andesite MI – QFM

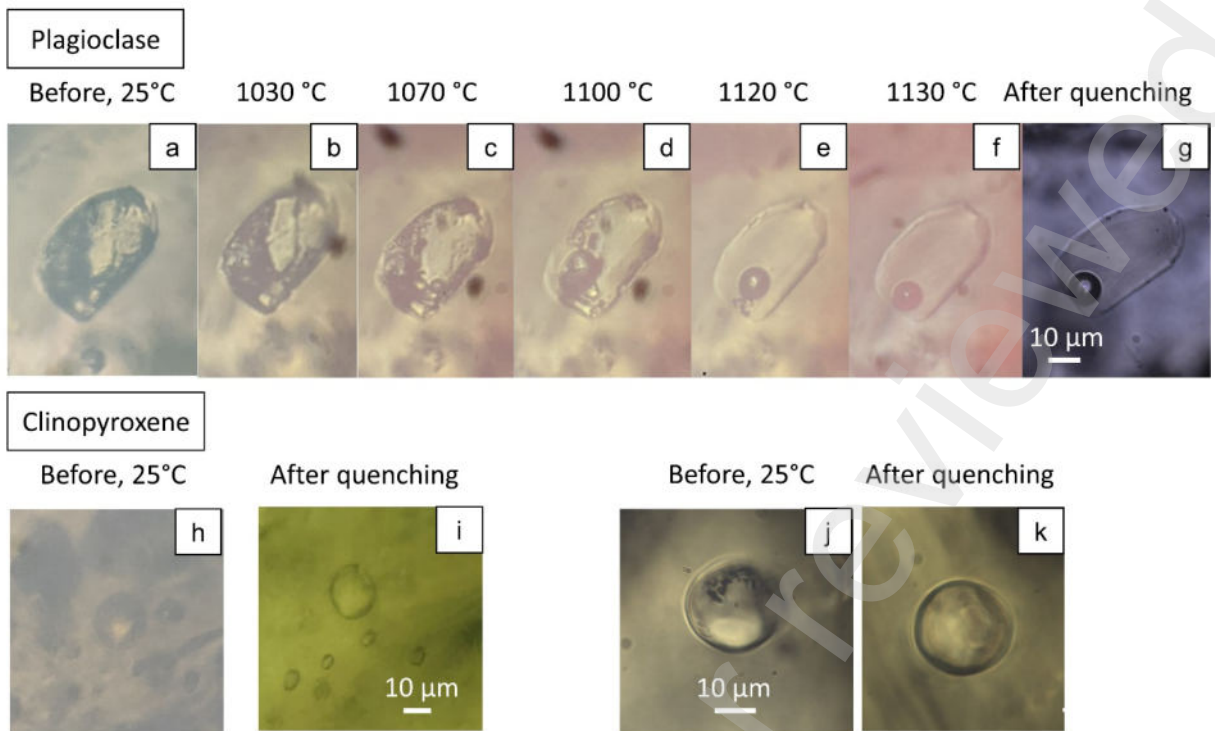
Literature data on Iran

- ⊠ Alborz Paleogene lavas (Verdel et al., 2011)
- ⊠ Pliocene-Quaternary Qorveh-Bijar belt lavas (Salehi et al., 2020)
- Pliocene-Quaternary Qorveh-Bijar belt olivines-hosted MI (Salehi et al., 2020)

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1236 **Figure 2**

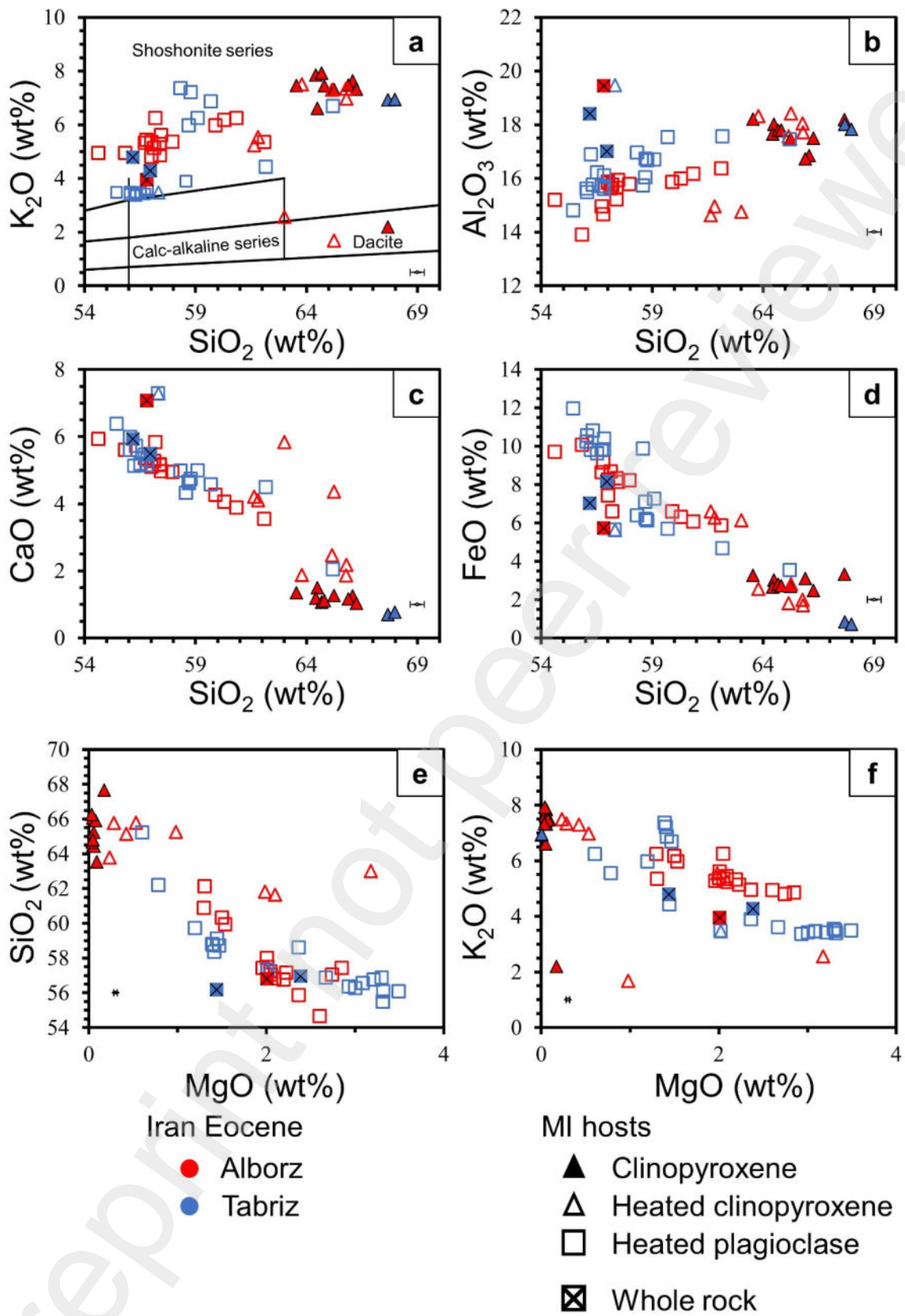
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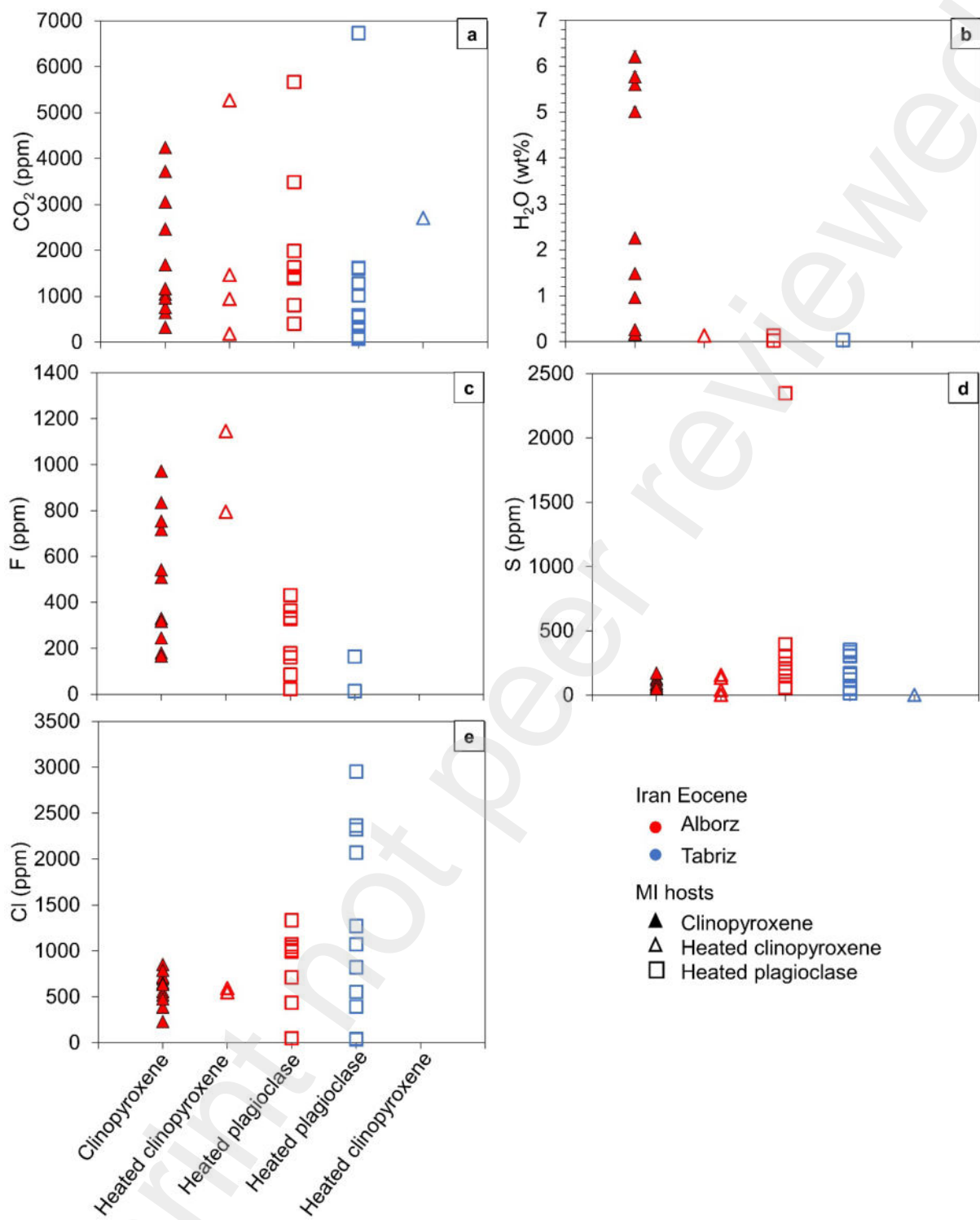
1239 **Figure 3**

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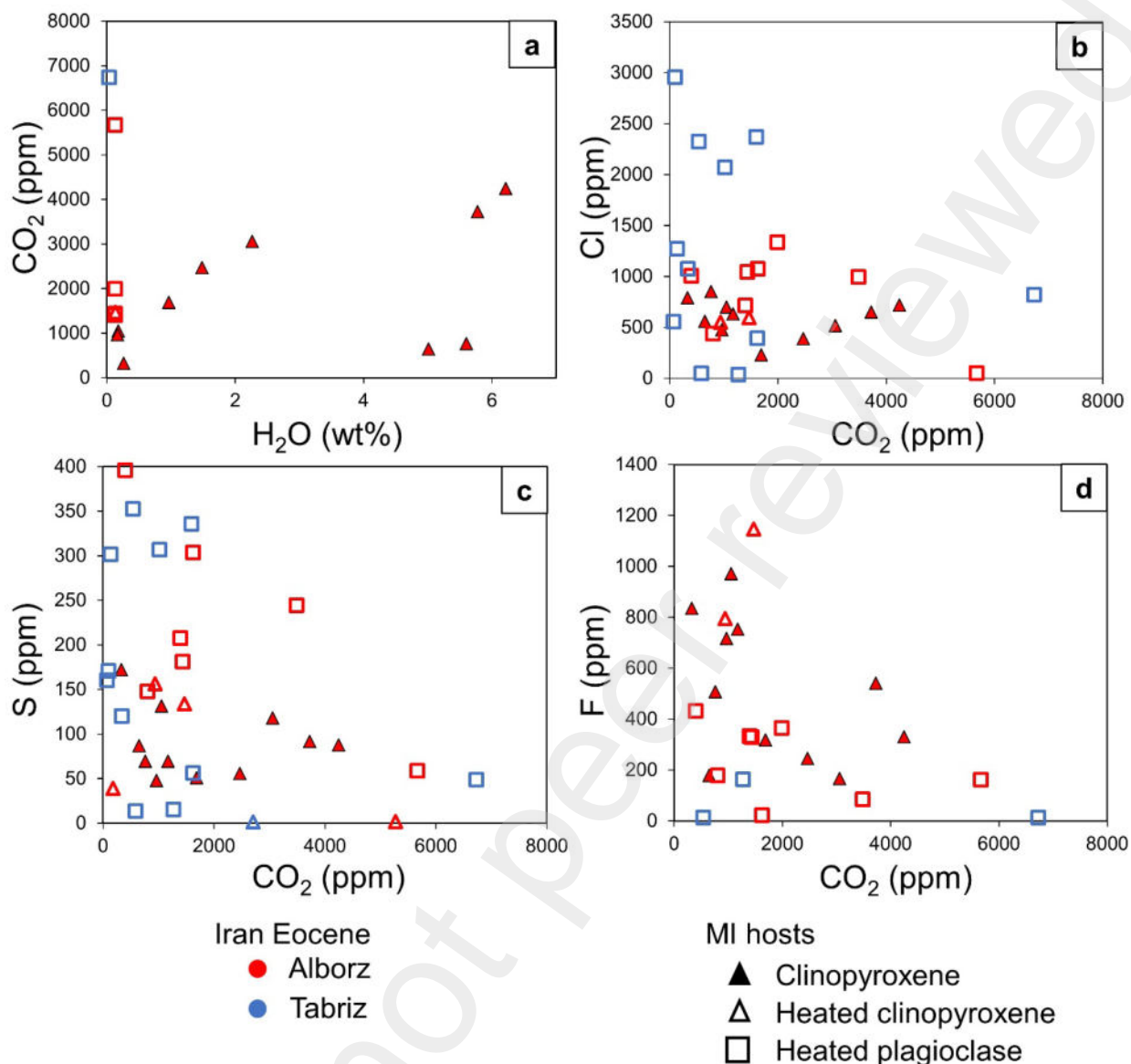
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1242 **Figure 4**



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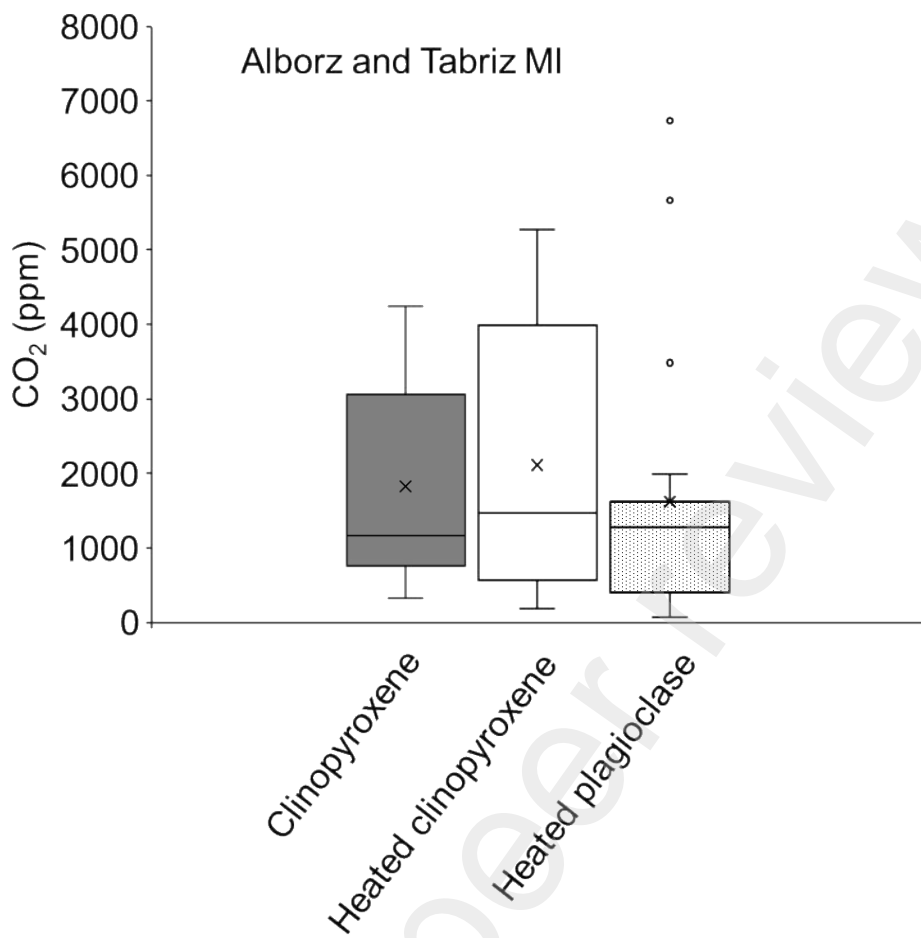
1244 **Figure 5**



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1246 **Figure 6**

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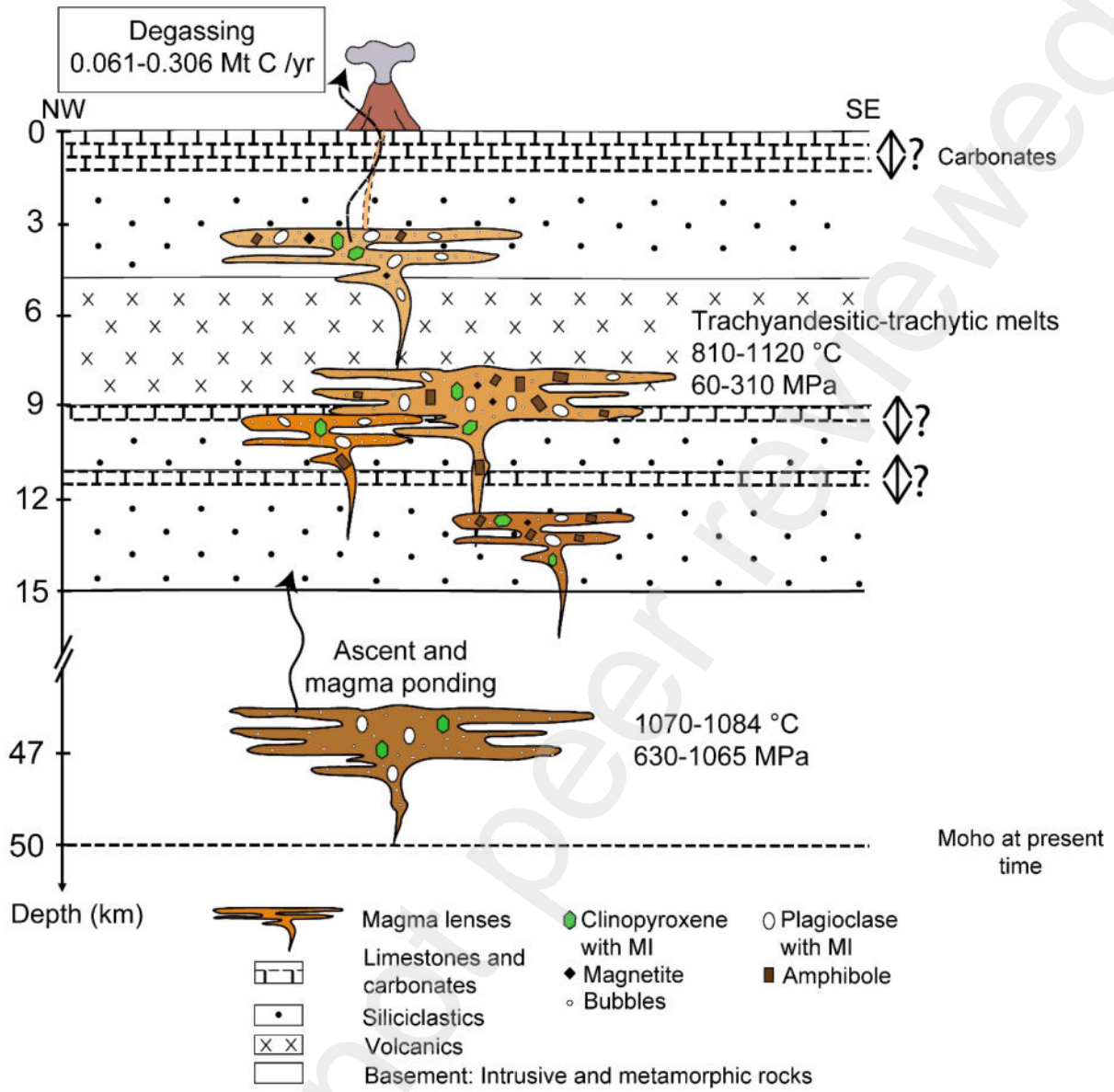
1249 **Figure 7**

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1255 **Figure 8**

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1257 **Tables**

Sample	Location	Coordinates	Nature of the sample	Approximate age
E1a, E1c	Khoja	38° 26.041'N, 46° 55.037'E	Trachyandesite	Middle-late Eocene (Geological Survey and Mineral Exploration of Iran, 2006)
PS2	Karaj Formation, Alborz Mountains	35° 57.970'N, 50° 52.932'E	Trachyandesite	Middle-late Eocene (Geological Survey and Mineral Exploration of Iran, 1995)

1258 **Table 1: Description of the samples from Iran studied here (location, coordinates (in the**
 1259 **World Geodetic System WGS84) and nature of the samples are specified).**

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Temperature (°C)	Heating rate – duration during each temperature plateau	Crystal type
25-300 300-700 700-900 900-1000 1000-1100 1100-1120 1120-1130 1130-1135 1135-1140 1140-1145 1145-1150 1150-1155 1155-1160	100 °C/min – 3 min 100 °C/min – 3 min 100 °C/min – 10 min 50 °C/min – 15 min 10 °C/min – 35 min 1 °C/min – 15 min 1 °C/min – 15 min 1 °C/min – 15 min 1 °C/min – 15 min 1 °C/min – 15 min 1 °C/min – 15 min 1 °C/min – 10 min 1 °C/min – 10 min	Clinopyroxenes
25-300 300-700 700-900 900-1000 1000-1100 1100-1120 1120-1125 1125-1130 1130-1135	100 °C/min – 3 min 100 °C/min – 3 min 100 °C/min – 10 min 50 °C/min – 15 min 10 °C/min – 35 min 1 °C/min – 15 min 1 °C/min – 15 min 1 °C/min – 20 min 1 °C/min – 15 min	Feldspars

1261 **Table 2: Heating ramps used for experiments on crystals from Iran. The durations of**
 1262 **maintaining the temperature constant after each temperature step are also specified.**

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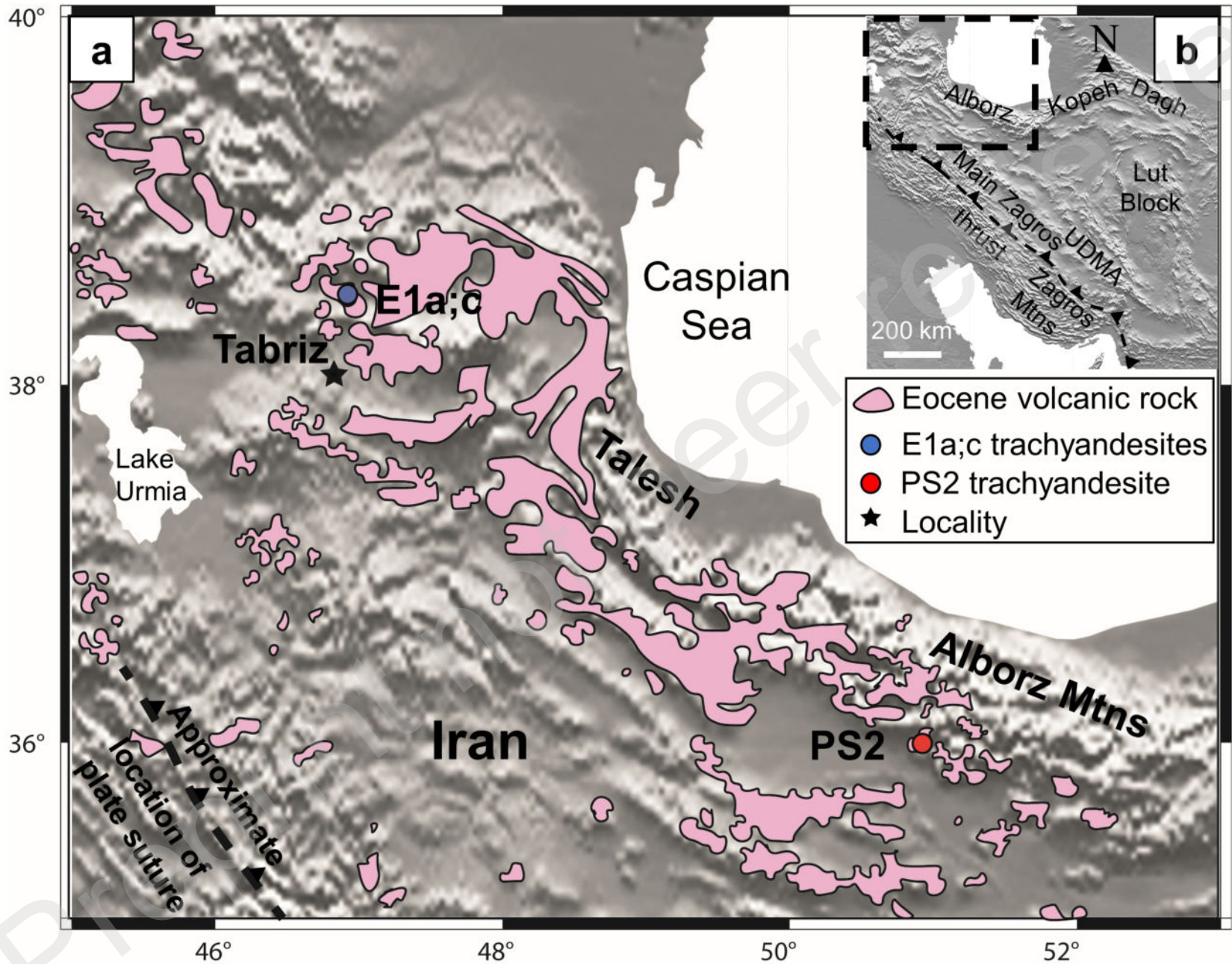
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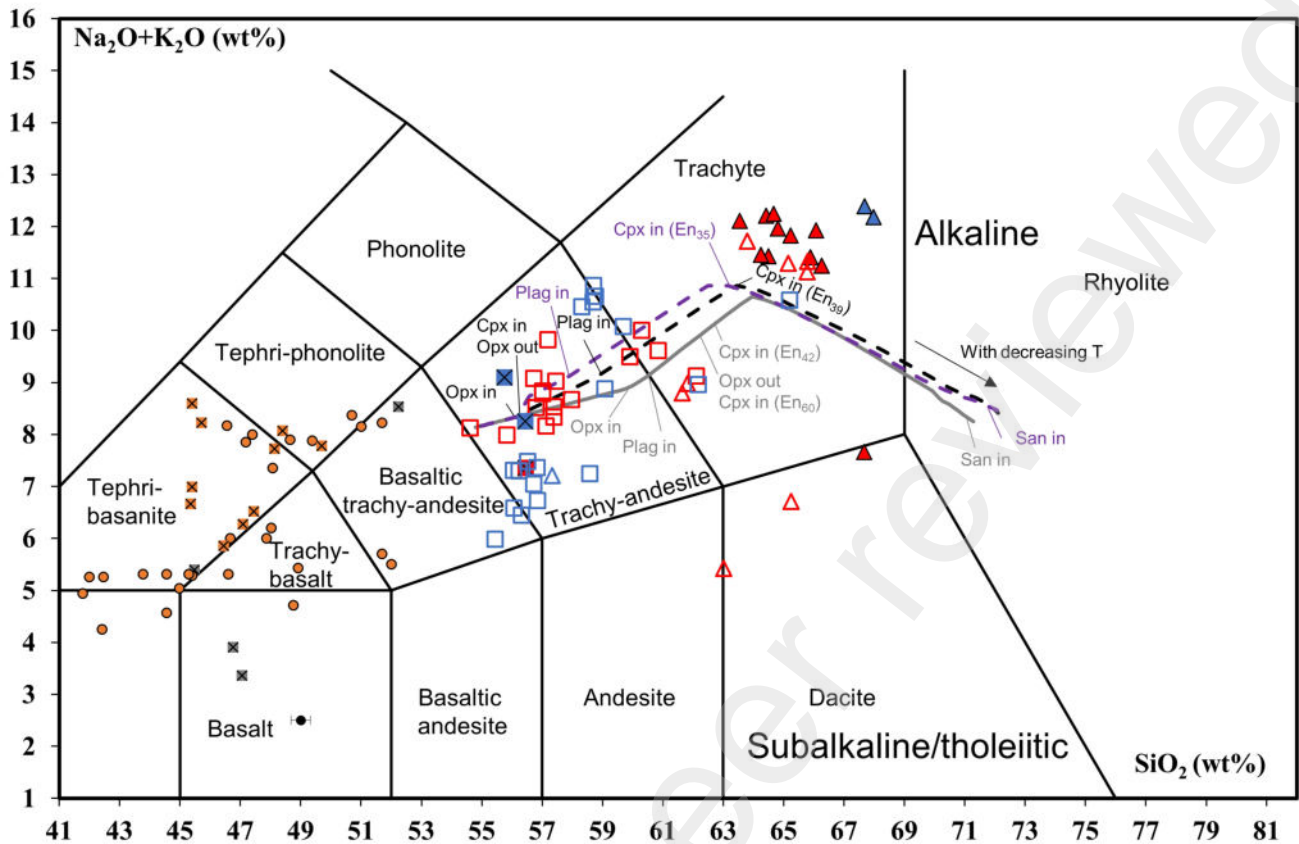
Sample	PS2 (Alborz)	E1a (Tabriz)	E1c (Tabriz)
Nature	Trachyandesite	Trachyandesite	Trachyandesite
SiO ₂ (wt%)	54.88	53.54	54.67
TiO ₂ (wt%)	0.92	0.94	0.95
Al ₂ O ₃ (wt%)	18.78	17.54	16.33
Fe ₂ O ₃ (wt%)	6.15	7.44	8.71
MnO (wt%)	0.1	0.2	0.16
MgO (wt%)	1.94	1.37	2.29
CaO (wt%)	6.83	5.65	5.27
Na ₂ O (wt%)	3.33	4.17	3.88
K ₂ O (wt%)	3.82	4.57	4.11
P ₂ O ₅ (wt%)	0.47	0.62	0.49
Cr ₂ O ₃ (wt%)	0.023	<0.002	<0.002
Ba (ppm)	639	674	630
Ni (ppm)	<20	<20	<20
Sr (ppm)	610	502	450
Zr (ppm)	219	203	191
Y (ppm)	26	29	31
Nb (ppm)	15	11	10
Sc (ppm)	14	14	16
LOI (wt%)	2.5	3.7	2.8
Total	99.88	99.89	99.85

1291 **Table 3: Whole rock major and trace elements of the Iranian samples from Alborz and**
1292 **Tabriz. The sampling locations are specified in Table 1. Analyses have been performed**
1293 **at Bureau Veritas Mineral Laboratories (Canada) by ICP-OES (Inductively Coupled**
1294 **Plasma-Optical Emission Spectrometry). The type of deposit is specified. Data are**
1295 **expressed with total Fe as Fe³⁺ with LOI detailed.**

Iran						Relative uncertainty (%)
Volume emitted over 9 My (45-36 Myr) (van der Boon, 2021) (km ³)	Minimum		Maximum			
	7.6×10 ⁴		3.8×10 ⁵			
Average density Iran magmas (g.cm ⁻³) (using the Iqpet software; Carr and Gazel, 2017)	2.32					
Total erupted mass over the last eruption (t)	Minimum		Maximum			
	1.76×10 ¹⁴		8.82×10 ¹⁴			
Average fractionation (%) (Macdonald et al., 2008)	70					
Total mass of parental magma (t) taking the minimum and maximum volume erupted over the last eruption	Minimum		Maximum			
	3×10 ¹⁴		1.5×10 ¹⁵			
Measured CO ₂ content from SIMS (ppm)	Minimum	Maximum	Minimum	Maximum	Average	
	76	6733	76	6733	1757	1
Calculated CO ₂ flux (t/d) assuming CO ₂ contents of MI from Iran	9.25×10 ⁵	8.19×10 ⁷	4.62×10 ⁶	4.10×10 ⁸	1.07×10 ⁸ (using the maximum mass of magma)	4
Total emission (g CO ₂)	2.28×10 ¹⁶	2.02×10 ¹⁸	1.14×10 ¹⁷	1.01×10 ¹⁹	2.63×10 ¹⁸	4
CO ₂ flux (Mt CO ₂ /yr)	0.003	0.22	0.01	1.12	0.29	4
Total emission (g C)	6.22×10 ¹⁵	5.51×10 ¹⁷	3.11×10 ¹⁶	2.75×10 ¹⁸	7.19×10 ¹⁷	4
Carbon flux (Mt C/yr)	0.001	0.06	0.003	0.306	0.080	4

1296 **Table 4: Magmatic CO₂ flux mass balance calculations for Iran magmas assuming CO₂**
1297 **concentrations of MI by SIMS. CO₂ values could be higher considering CO₂ released from**
1298 **carbonate sediments.**





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- ⊠ Whole rock

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- Basaltic trachy-andesite MI – no buffer
- - Basaltic trachy-andesite MI – NNO
- - - Basaltic trachy-andesite MI – QFM

Literature data on Iran

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- ⊠ Pliocene-Quaternary Qorveh-Bijar belt lavas (Salehi et al., 2020)
- Pliocene-Quaternary Qorveh-Bijar belt olivines-hosted MI (Salehi et al., 2020)

Plagioclase

Before, 25°C

1030 °C

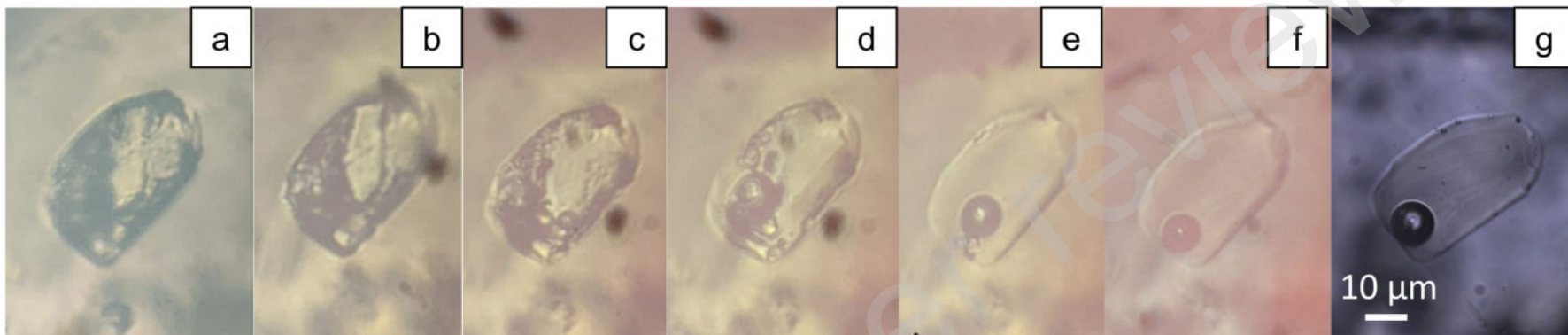
1070 °C

1100 °C

1120 °C

1130 °C

After quenching



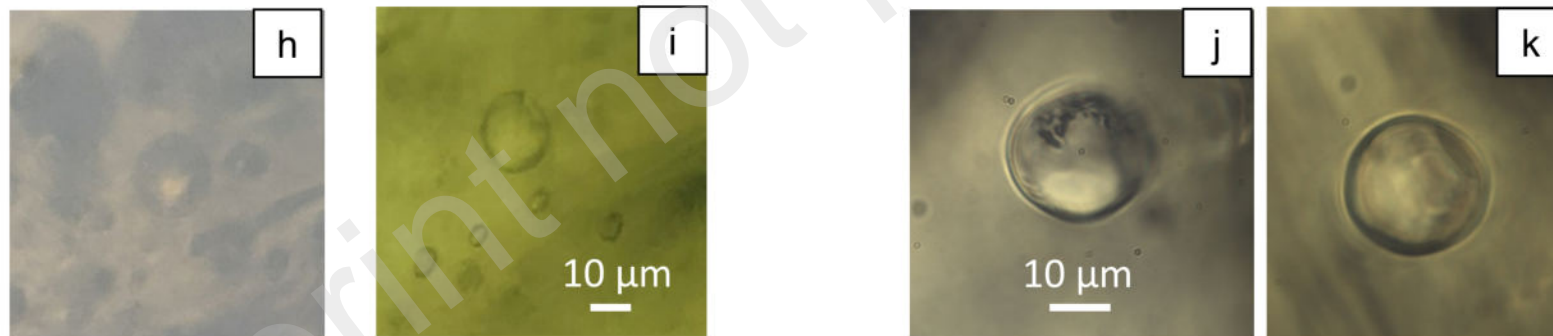
Clinopyroxene

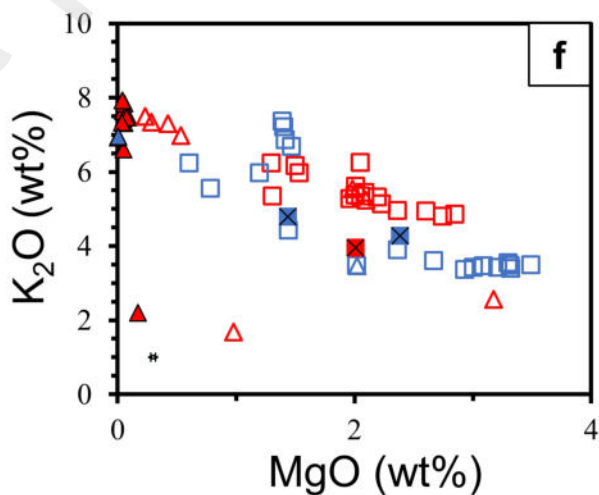
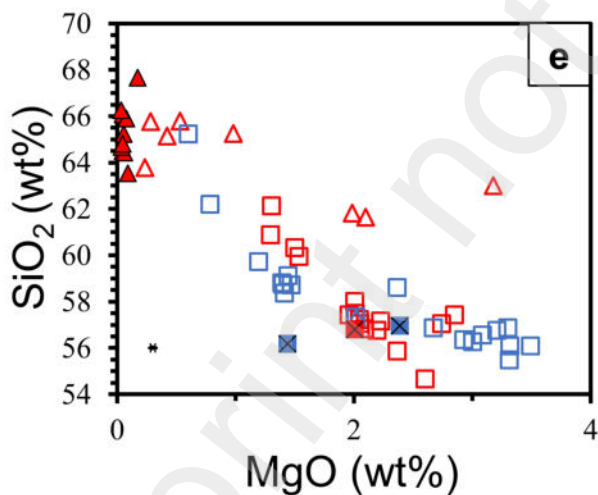
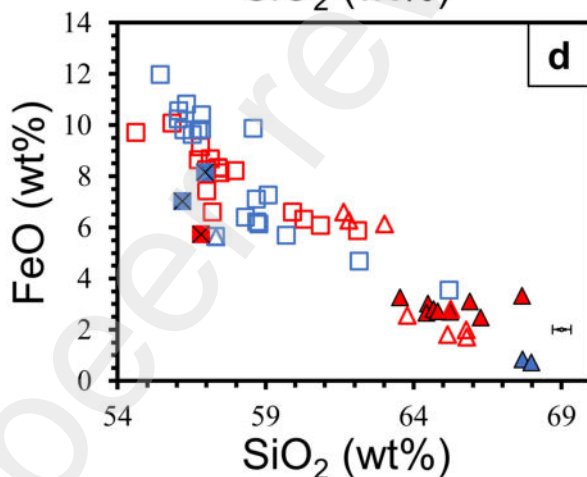
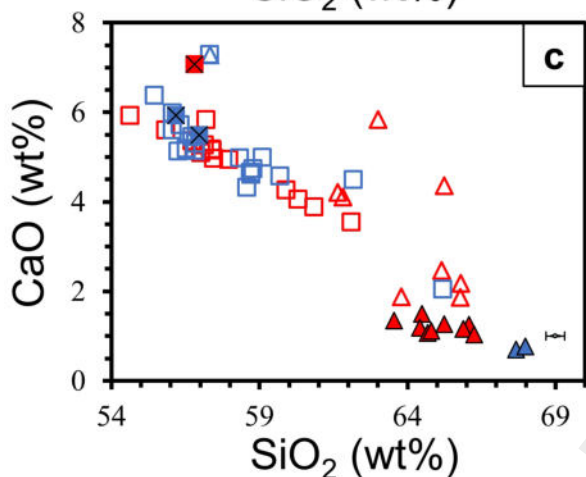
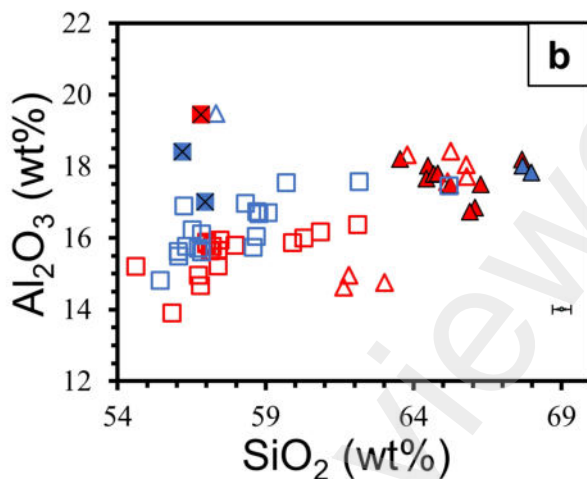
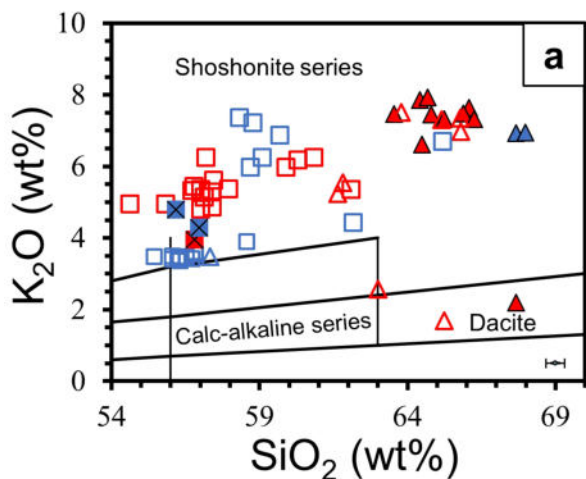
Before, 25°C

After quenching

Before, 25°C

After quenching



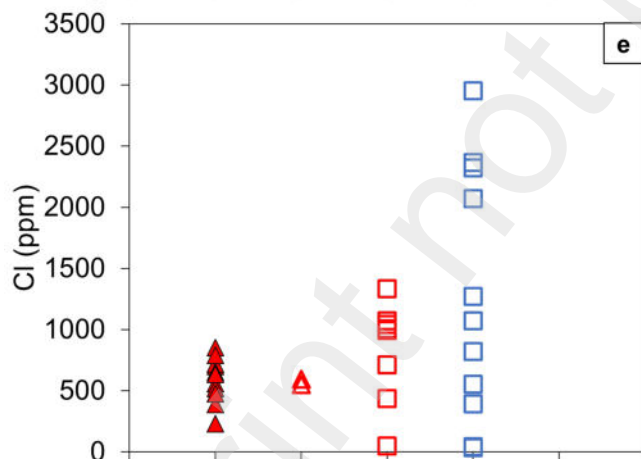
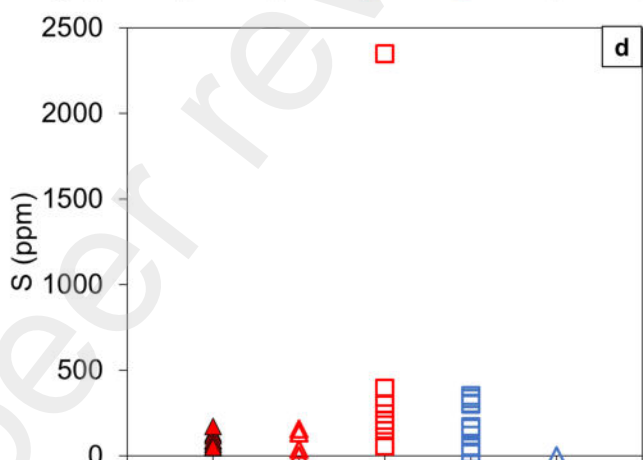
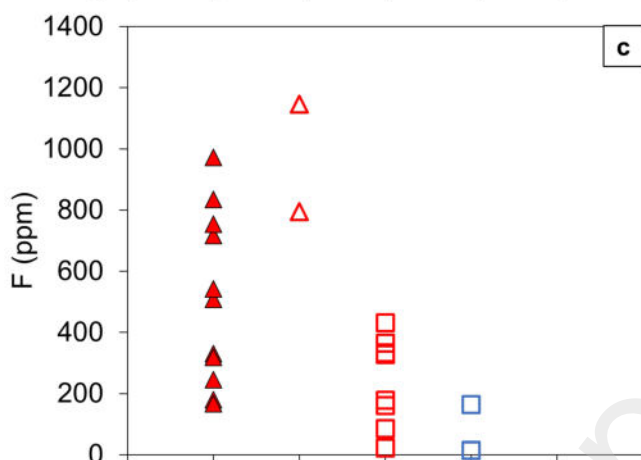
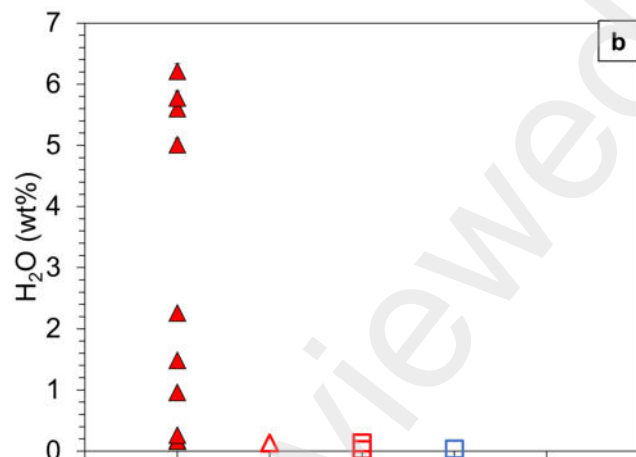
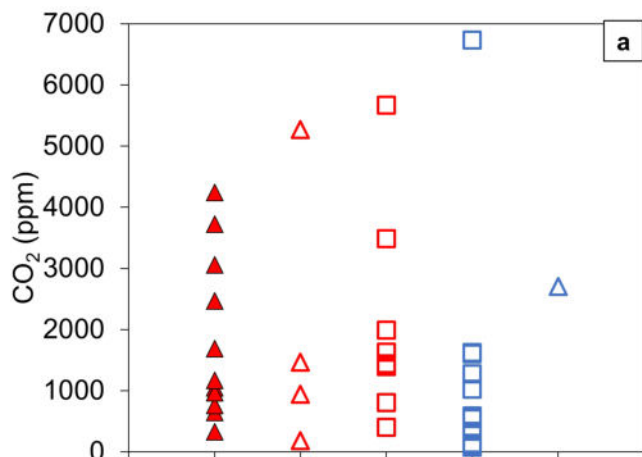


Iran Eocene

- Alborz
- Tabriz

MI hosts

- ▲ Clinopyroxene
- △ Heated clinopyroxene
- Heated plagioclase
- ⊗ Whole rock



Iran Eocene

● Alborz

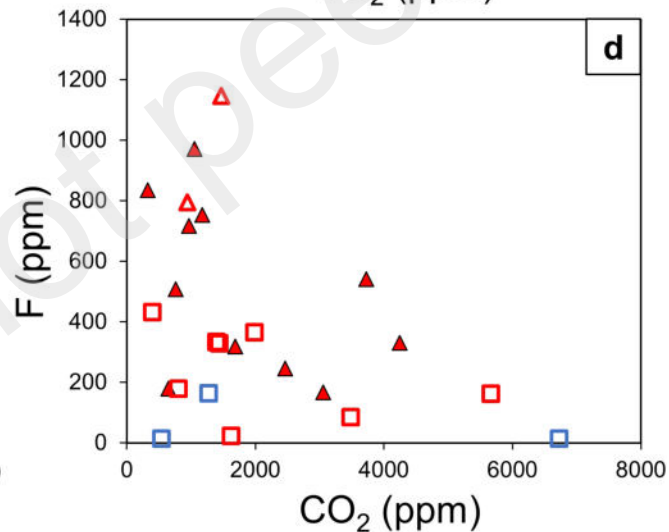
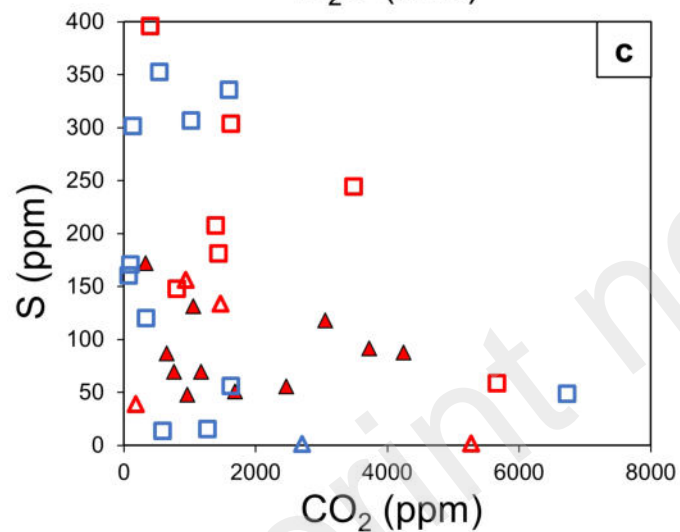
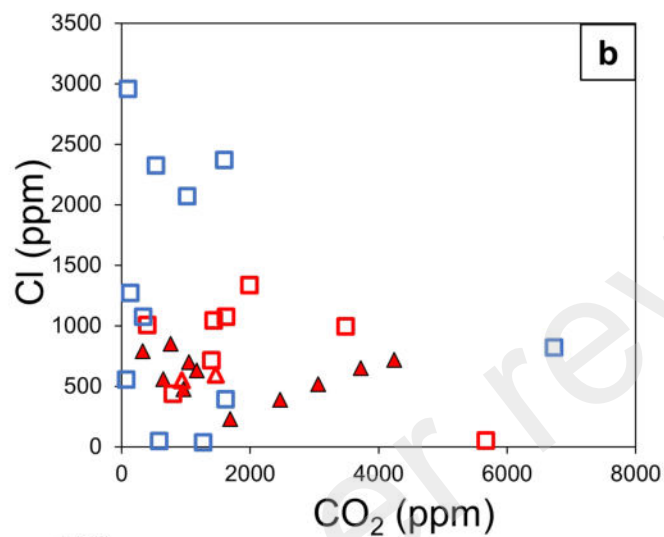
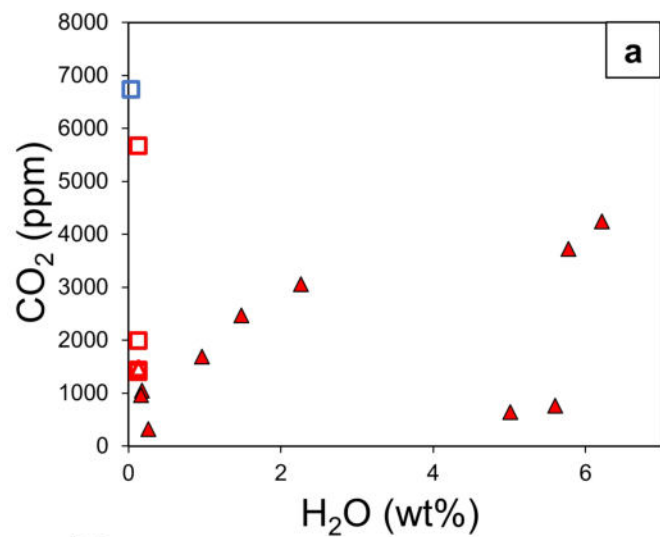
● Tabriz

MI hosts

▲ Clinopyroxene

△ Heated clinopyroxene

□ Heated plagioclase



Iran Eocene

● Alborz

● Tabriz

MI hosts

▲ Clinopyroxene

△ Heated clinopyroxene

□ Heated plagioclase

Alborz and Tabriz MI

